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**LETTER REPORT  
GARFIELD ALLOYS FIRE SITE  
GARFIELD HEIGHTS, CUYAHOGA COUNTY, OHIO**

**Prepared for:**

**U.S. ENVIRONMENTAL PROTECTION AGENCY  
Region 5 Emergency Response Branch  
9311 Groh Road  
Grosse Ile, Michigan 48138**

TDD No.:	S05-0312-007
Date Prepared:	March 30, 2004
Contract No.:	68-W-00-129
Prepared by:	Tetra Tech EM Inc.
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## **Tetra Tech EM Inc.**

6801 Engle Road, Suite G ♦ Cleveland, Ohio 44130 ♦ (440) 234-0886 ♦ FAX (440) 234-1725

March 30, 2004

Mr. Jeffrey Kimble  
On-Scene Coordinator  
Emergency Response Section #1  
U.S. Environmental Protection Agency  
9311 Groh Road  
Grosse Ile, Michigan 48138

**Subject: Letter Report**  
**Garfield Alloys Fire Site**  
**Garfield Heights, Cuyahoga County, Ohio**  
**Technical Direction Document No. S05-0312-007**  
**Tetra Tech Contract No. 68-W-00129**

Dear Mr Kimble:

The Tetra Tech EM Inc. (Tetra Tech) Superfund Technical Assessment and Response Team (START) prepared this letter report in accordance with the requirements of Technical Direction Document (TDD) No. S05-0312-007 issued by the U.S. Environmental Protection Agency (U.S. EPA). The scope of this TDD was to conduct emergency response activities at the Garfield Alloys Fire (Garfield Alloys) Site in Garfield Heights, Cuyahoga County, Ohio. Specifically, Tetra Tech was tasked to prepare a health and safety plan, document site conditions with written logbook notes and photographs, conduct air monitoring, conduct multimedia sampling and make recommendations to the U.S. EPA based on site conditions and monitoring and sampling results. Emergency response activities were conducted by START members Stephen Wolfe, Kelly Smith, and Anne Busher. This report discusses site background information, emergency response activities, and analytical results.

The Garfield Alloys Site is the location of a massive fire that started in and engulfed Garfield Alloys, Inc., a magnesium recycling facility and a number of other surrounding buildings. The site is located at 4878 Chaincraft Road, Garfield Heights, Cuyahoga County, Ohio. The geological coordinates are 41°25.759' North and 81°35.803' West. The site occupies 16 acres and includes four buildings. The recycling facility processes and recycles magnesium into ingots for resale. The site is bordered to the north by the Norfolk Western Rail Road Line and a large cemetery; to the west by the Garfield Park Reservation and industrial areas; to the south by Chaincraft Road, Mill Creek, the Garfield Park Reservation and residential areas; and to the east by an industrial and residential area.

The fire started at approximately 3:00 p.m. on December 29, 2003, inside an area where 55-

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gallon drums of magnesium arrived from other facilities and were opened. At the time of the incident, the company estimated that approximately 1,000,000 pounds of magnesium was present. Employees initially attempted to extinguish the fire, but it soon grew out of control. The Garfield Heights Fire Department responded to the fire and set up an incident command. At 5:00 p.m., U.S. EPA was requested to assist with the situation. At 6:00 p.m., U.S. EPA tasked Tetra Tech to respond to the fire and provide assistance to the U.S. EPA (see Attachment A, Log Book).

Magnesium is easily ignited and highly reactive with moisture; therefore, the rainy weather conditions on December 29 significantly complicated firefighting efforts. The large plume of smoke, flashes, sparks, and bright white light from the fire could be seen for miles (see Attachment B, Photographic Log). The loud explosions rattled the ground, even breaking windows in a nearby apartment complex. A product of the magnesium fire and the reaction of magnesium with moisture is magnesium oxide, a respiratory and eye irritant.

Tetra Tech arrived at the site at approximately 7:30 p.m. Routine air monitoring for volatile organic compounds (VOC), combustible gases, and oxygen was conducted at eight residential and industrial locations around the fire (see Attachment C, Air Monitoring and Sampling Locations Map and REAC DataRAM Results). Particulate and chemical-specific monitoring (for chlorine, ammonia and acid gases) was also conducted during the night and into the next days with a DataRAM and colorimetric Draeger tubes. VOC monitoring was conducted using a flame ionization and photoionization detectors (FID and PID). FID and PID monitoring results for VOCs mostly ranged from 0.0 to 2.70 parts per million (ppm), which did not exceed background levels. One result of 4.12 ppm was due to vehicle exhaust. Air monitoring results for radiation, oxygen, hydrogen sulfide, combustible gases, chlorine, ammonia, and acid gases were either nondetect or did not exceed background levels (see Attachment D, START Air Monitoring Log). START DataRAM results ranged from 0.002 to 2.6 milligrams per cubic meter ( $\text{mg}/\text{m}^3$ ) in the middle of the densest portion of the smoke plume.

The City of Cleveland Fire Department mobilized a HAPSITE™ portable field gas chromatograph/mass spectrometer and an operator to the scene to provide real-time analytical data for VOCs from air samples collected by Tetra Tech. The air samples were collected in Tedlar bags at several locations during the late hours of December 29 and the early hours of December 30, 2003 including locations downwind of the plume and in residential areas. No VOCs were detected in any of the samples.

Tetra Tech collected a SUMMA canister of air from the smoke plume (sampling location #3) for volatile analysis (Method TO-14). In addition, two absorbent tube air samples for metals analysis and one absorbent tube sample for a VOC and petroleum hydrocarbon scan were

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collected from a residential neighborhood (sampling location #6). An additional absorbent tube air sample for metals analysis was collected from a downwind industrial location (sampling location #4), and another tube sample was collected from a residential area (sampling location #7). Absorbent tube air samples were delivered early the morning of December 30, 2003, to the laboratory for quick-turnaround analysis for magnesium oxide and a VOC and petroleum hydrocarbon scan. At 4:00 p.m. on December 30, 2003, draft analytical results for magnesium oxide reported by the laboratory ranged from nondetect to  $0.84 \text{ mg/m}^3$ . The VOC and petroleum hydrocarbons scan results were reported as nondetect (see Attachment E, START Analytical Data Results).

U.S. EPA, the Cuyahoga County Health Department, and the Ohio Department of Health established the magnesium oxide action level for residential areas as  $10 \text{ mg/m}^3$ . The Agency for Toxic Substances and Disease Registry concurred that this level was appropriate for short-term exposure. This action level for magnesium oxide is based on the 8-hour, time weighted average (TWA) of  $10 \text{ mg/m}^3$  for occupational exposure.

On December 30, 2003, the U.S. EPA Environmental Response Team (ERT), U.S. EPA Response, Engineering, and Analytical Contract (REAC) contractor, and START contractor Weston Solutions (Weston) arrived on site. REAC was tasked to collect 12 8-hour air samples (see Attachment C) for total metals analysis from locations immediately adjacent to the burn area and from the site perimeter. In addition, REAC set up three DataRAM particulate monitors around the burn area to assess particulate concentrations over an 8-hour period (see Attachment C). Weston assisted with conducting routine air monitoring at the eight residential and industrial locations around the fire. Weston also collected global positioning system satellite information for each of the eight sampling locations.

On December 30, U.S. EPA requested the mobilization of the ASPECT response aircraft from Region 7 to conduct thermal imaging and aerial photography, and to provide a Fourier transform infrared (FTIR) scan for the magnesium oxide concentrations in the smoke plume and at the fire site. The aircraft conducted eight passes over the fire with similar results. FTIR scan results indicated that a plume of the magnesium oxide was not detected leaving the fire at the time of the flyover. Specific magnesium oxide results could not be obtained from the FTIR scan, possibly because of the high energy radiance from the fire and the large amount of heat produced (see Attachment F, ASPECT Report).

On December 31, 2003, REAC completed the 8-hour air sampling event and sent the samples to the U.S. EPA ERT laboratory in Edison, New Jersey for metals analysis. The SUMMA canister sample collected by Tetra Tech was also sent to the ERT laboratory for VOC analysis. A representative from the U.S. Chemical Safety and Hazard Investigation Board and two State of

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Ohio Fire Marshals were on site to conduct site and fire investigations.

VOC Results from the SUMMA canister air sample collected on December 30, 2004, were reported as nondetect for all compounds (see Attachment G, SUMMA Canister Analytical Results). The results for the 11 air samples collected (1 sample was lost) by REAC on December 31, 2003, and analyzed for total metals are presented Attachment H, REAC Air Sample Results.

If you have any question or comments regarding this deliverable, please contact me at (440) 234-0886 ext. 225, or Tom Kouris at (312) 946-6431.

Sincerely,



Anne A. Busher  
START Project Manager  
Tetra Tech EM Inc.

Attachments:

- A Log Book
- B Photographic Log
- C Air Monitoring and Sampling Locations Map and REAC DataRAM Results
- D START Air Monitoring Log
- E START Analytical Data Results
- F ASPECT Report
- G SUMMA Canister Analytical Results
- H REAC Air Sample Results

cc: Lorraine Kosik, U.S. EPA START Project Officer  
Thomas Kouris, Tetra Tech EMI START Program Manager Letter Report Information

Attachment A

Log Book

"*Rite in the Rain*"  
ALL-WEATHER WRITING PAPER



## LEVEL

All-Weather Notebook  
No. 311

Garfield Allys
4070 Chaucer Rd.
Garfield Hts, OH
TWO: 505-0912-007

4 5/8" x 7" - 48 Numbered Pages

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Jim Iwata  
216.789.9281



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"Rite in the Rain"  
ALL-WEATHER WRITING PAPER



Name Anne Bushee - TetraTech  
START Contractor

Address Kelly Smith - TetraTech  
START Contractor

Phone \_\_\_\_\_

Project OSC - Jeff Kimble  
U.S. EPA Grobe Ilc, m.

Clear Vinyl Protective Slipcovers (Item No. 30) are available for this style of notebook.  
Helps protect your notebook from wear & tear. Contact your dealer or the J. L. Darling Corporation.

12/29/03

# garfield Alloys Fire

## CONTENTS

TIME

REFERENCE

DATE

1800 START Busher contacted by  
OSC Brad Stimpert of U.S. EPA  
to respond to magnesium fire  
in Garfield Hts, OH.

1815 left house for office &  
contacted Steve Wolfe  
to respond to.

1830 Arrived at the office  
to begin HAZOP.

1845 START Wolfe arrived  
& began to load up the vehicle  
w/ equipment.

1900 Researched a little on  
magnesium fires.

1910 left the office for  
the site. Talked with  
OEPA OSC Jim Irwin on site.  
he will meet us at the  
Garfield Hts, Fire Dept.  
and we will go to site.

1930 Arrived on site and were  
met by OSC Stimpert.

Amak 12/29/03

12/21/03 Garfield Alloys  
2000 Drove down to the site  
to see the situation and  
talk to the chief.  
OSC Irwin + Stimpfe talked to  
Garfield Hts. Fire Chief  
~~Collopy~~ ~~He~~ He would like us  
to monitor for volatiles, n<sub>2</sub> and  
and LEL, O<sub>2</sub> in the plume  
in the residential areas  
along Miles Rd. and E 93rd St  
2130 Continued to watch the  
fire as OSC Irwin and Stimpfe  
continue to collect info.  
2200 Drove along miles w/ OSC  
Irwin + OSC Kinble, who was  
having arriving at site.  
Arrived at a school along miles at  
E. 115th St. to calibrate the equipment.  
to monitor the air in the residential  
neighborhoods.

Amak

12/21/03

12/25/03

Sanfield Alloys

2230 Complete calibration of instruments.

Pump # ~~14600~~ 14610

calibrated 1.022 Liter/min

sample for metals MT-Φ1

calibrated TVA 1000 model ~~14600~~

70547 to zero air

calibration ok; battery ok

calibrated the PID w/ isobutylene

and the FID w/ methane

calibrated the CGE to

O<sub>2</sub> and CEL and H<sub>2</sub>S

all ready to go

background with TVA 1000

is 2.63 ppm

Pump # 14650 calibrated to

1.109 liter/min; sample

for metals. MT-Φ2

AmV 12/29/03

12/25/03 Sanfield Alloys  
Pump # Cleveland 3  
1,706 liters/min will be  
run for volatiles

---

2307 Sample VO-01; volatilize  
organic sample started.  
Sample collected off of  
Miles ave. between E. 111<sup>th</sup> and  
E. 110<sup>th</sup> streets.  
Pump # MT-01

2309 13 Sample VO-02 MS  
set up on; sample for metal  
pump number 1465B

---

~~2308 MS~~ 2310 Sample for metals  
MT-02 with sample pump  
number 14660 was set.  
The sample was set up  
along the fence line along  
of ~~and~~ Miles Road between 111<sup>th</sup>  
and E. 112<sup>th</sup> streets along  
a back

12/29/03

- 12/25/03 Sanfield Alloys
- 2314 Pump time for the Volatiles  
VO-01-092 7 minutes; Pump #3
  - 1.755 liters/min - final calibration
  - Readings w/ TVA-1000 were  
3.73 ppm; background  
was 2.163 ppm
  - Weather: temp 40°F; rain;  
wind out of SW; temps  
dropping.
  - Readings w/ the PHP-1.1k  
were O<sub>2</sub> 20.3 % O<sub>2</sub>;  
LEL = 0; and CO = 0.0
  - Reading w/ the U.S. EPA multi  
RAE SN = PG-150-SF =  
VOC = 0.0; CO = 0.0; H<sub>2</sub>S =  
0; O<sub>2</sub> = 19.6%; and  
LEL = 0.0.

2323 readings on the TVA1000

- = 4.12 ppm
- Collected the Draeger ammonia,  
chlorine + acid tubes all were  
non-detect. SN 12/25/03

12/29/03 Garfield Allogo  
2330 readings on the TUA in  
the plane were 3.17 ppm

2335 M $\phi$ - $\phi$ 1 sample was collect  
final cal = 1.033 L/min at the fence  
btwn E. 110 + E. 111 streets

2340 M $\phi$ - $\phi$ 2 sample was collect  
final cal = 1.073 L/min at the fence  
btwn E. 110 + E. 111 streets

00114604 E. 131st Garfield garden's  
Apt. collected 1 Tedlar

air-sample for Cleveland Fire

Dept. in air sampler

Bring it back to the Garfield

HHS Fire Dept. to run it on


the HAP site that belongs to

Cleveland Fire Dept. Terry

Bindermaegel is here to operate

the HAP site.

The sample was collected at  
the entrance to an apartment  
complex. In a residential  
neighborhood.

 12/29/03

12/30/03

Sanfield Alloys

Sample Resident - 01 at 4604 E. 131st  
Street.

150 14200 Broadway Ave.

collected 1 Tedlar bag

and 1 Shuman canister of  
air from underneath the

major portion of the plug

0145 Returned to the Fire dept.

so that Lt. Biedermeier could run

the sample on the HAPSITE.

Readings taken with the TUA 1000

were 2.50 ppm on the FID

the PIP is not operating correctly,  
possibly due to rain/humidity.

COI was 20.3% O<sub>2</sub>; LEL

was 0.0%; H<sub>2</sub>O = 0.0 ppm;

0200 HAPSITE readings are

not above background. It is

a standard chart.



AK

~~12/30/03~~  
12/30/03



12/30/03 Garfield Hhs  
Earlier in the evening is  
the fire significantly expanded  
when the fire spread to a second  
building. The explosions rattled the  
doors of the Garfield Hhs fire dept  
located at Turney and McCracken.  
The sounds of the explosions occur  
one after another - in quick  
succession. The explosions lit up the  
sky and you could see sparks high  
up in the clouds. The clouds  
generated by the explosions  
filled the sky.

The Hhs site was staged in the  
Garfield Hhs, Fire Department.  
We - START - ran samples back &  
forth.

Weather: weather continues to  
be raining, temps have dropped to  
low 40's - upper 30's; winds have  
shifted the plume towards  
the intersection of McCracken  
Road and Turney Rd. Rain is  
hard + steady.

CR 12/30/03

12/30/03 Garfield Allego  
Fire Fighters mention that the  
area we first assembled in was  
now flooded.

OEPA & the U.S. EPA contact the plant  
managers. Greg OKR contacted the  
plant manager to get someone out  
here to monitor the water quality  
of the creek. Slovic, Bellevue,  
Mike & Chuck Stotski, Bellevue,  
in Magnatech Company.

Sample location #1 Marymount  
Hospital at the 1st drive off  
Mc Cracken

#2 Clarendon & Park Knoll

#3 Mc Cracken & Broadway  
in the Tonite Lounge parking lot

#4 3rd St & Broadway

#5 6131 of Street 46004

#6 Btw E. 110 & E. 111

#7 ~~St. Mary's~~ Langston

#8 Garfield & Broadway.

0430 Several hundred families  
have been evacuated.

SK 12/30/03

12/30/03 Garfield Alley

Chris - the health & safety person  
for Garfield Alley said that  
all we have is the magnesium  
on site.

Osc Kinsie and Kelly want to  
begin to make a plan

4775 - Kovak Automatic Sew Parts  
- diesel oil 5000 gal; lubricating oil 8-55 gal

\* 4825 Renew Furniture Refinisher

4865 Cardamone Construction  
2,000 heating oil

\* 4864 Clockwork Carrier

\* 4801 Chalcraft Taveis Container  
- 10 - 30lb propane tanks; flammable liquids  
in storage

\* no longer at that address  
12/30/03

- 12/30/03 Garfield Alloys  
\*4801 Chaiscraft VIP Insulation  
- bond master adhesive.

\*4801 Chaiscraft - in dust  
- none

\*4801 Chaiscraft grater landscaping  
Thompson Aluminum  
- 4  
- nothing

- 4850 Chaiscraft Thompson Aluminum  
- none

→ may have dust  
\*4801 Broadway Truck  
diesel + mineral spirits

- 4875 Custer Heat Treat Co.  
- 1-SS gal. diesel fuel.

\*4855 Halifax Ind.  
- none

\*4876 Chaiscraft  
@ GPB Performance Inc.

- vehicles inside  
GR 12/30/03

12/30/03 Garfield Alloy

Chuck Slouich owner

330-225-0022

Osc Kimble + Kelly call Mr. Chuck  
Slouich the owner of the Garfield  
Alloy

0557 Pump # 14658 started at 557  
0600 0.04 calibrated at 0557;

Will collect metals and be

WT-03 from Beachwood Clinic  
at Broadway + Forestview Rd.

START Busher calibrated the Person's

Note Room.

1.6 w/ wind and 0.45 w/out

wind. 2.10 in high smolder

We set the sample up at the Beach-  
wood Systems clinic located at  
Broadway. The wind is picking

up and there is visible smoke  
adjacent to this facility. is a  
trailer park w/ many trailers

A very light snow is falling. Temps  
have dropped to low 30's and

winds are out of NW going SE

multi RAC 0.0 ED; 0.0 H2S; 02 20.3°  
AD 0.0; TVA 0.0 H2S; 02 20.3°; FID = X FID;

12/30/03 MT-03

0620 Sample was collected and prepared

1.056 L/min was final calibration

Winds died down and smoke has reduced at present.

755 Pump # 14650 for sample MT-04 was initiated. at the corner of Edgepark and Layton in front of house  
1.063 L/min. Temp is low 30's; light snow

Reading for the Personal Data

Room was 0.0 mg/m<sup>3</sup>

The MultiRAE 0.0 H<sub>2</sub>S 20.8% O<sub>2</sub>  
0.0 CO Fl<sub>2</sub> = X

727 the sample is pulled off for analysis. The final flow rate is 1.060 L/min.

84 units were evacuated for the Henry Street Apartment. Another 80 units surrounding that area.

OK 12/30/03

- 12/30/03 Garfield Alloy  
1,000 tons of magnesium produced  
or 1, million pounds. The magnesium  
was in ingots; and what ever  
the product was. —

estimated that we used 0.25  
million gallons per hour of water.

0830 The Fire Dept. is putting a new product  
on the fire to see if it will stop it

1000 Determined that additional pump  
will be needed to assist.

1100 Requested that URS obtain  
a GIS person. to assist. —

1200 OSC Stimpale + OSC Dunn  
arrived on site; STARSS Smith +  
Biaglow arrived on site. to spell  
Wolfe and Busher

1400 Wester returns - very little plume  
left in air monitoring. Fire  
appears to be smoldering.

12/30/03. Garfield Alloy

1430 ERT REAC on site will set additional air sampled at

pre-determined location #1-8 and areas inside the boundaries of this area

Western completes 2nd Round of air monitoring

1700 GIS system update

OSC demob to pick up for flight-over

on site. REAC continues to collect samples. START has

completed Air monitoring plan;

and pickup to OSC

REAC: 10 sample locations

→ Cuyahoga County determined that the Action Level for residential  $MgO$  is not toxic but is an irritant for eyes & inhalation.

He stated that  $Mg$  is water reactive in its fine form.



He stated that GW & electricity have been turned off

Action Level  $10 \text{ mg}/\text{m}^3$  as  $MgO$ .



12/30/03 Garfield Alloys  
→ Transcribed notes from

Garfield Alloys

12205 Broadway, Garfield, OH

4878 Chaincraft, Garfield, OH

216-581-6355

- Talked to Chief Collova (Garfield Hts)
- Chief stated no citizens have called w/ complaints this far.
- He said no one was hurt or killed except fire dept personnel (from street).
- He was not sure how fire started
- Chief said he talked to a Greg (Facility Rep.) but Greg left NO info.
- Chief stated that the area was evacuated.
- ~~Jack Kerslake~~<sup>Jeff Kink</sup> & Brian Kelley, USEPA came in & reported findings from their sampling efforts.
- Chief called Cuyahoga County EMA. He stated that there was an industrial fire & that the Ohio & USEPA tested air quality.

12/30/03

## Garfield Alloys Fire

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- He stated that the area was evacuated and at the time

- There was no detection of contaminants in the air

- The Fire dept still recommended that citizen seal up their home

- & remain indoors w/ windows & doors closed.) If residents do

- leave there is a shelter at the Garfield fire station #1.

- A 1:46 AM Gorr & Jim Irwin went to Garfield fire dept.

- 2:05 G.orr calls Kauer (H & S Manager) 216-214-7804

- He said that the Bulk of the material was Mg (a couple of million pounds) in different forms:

① Mg flesh

② Mg Powder

③ Mg block

- There were also maintenance solvents, processing chemicals, 10 nitrogen cylinders & 6 SO<sub>2</sub> cylinders (2900 lbs)

12/30/03

Sanfield Alloys

Also Foundry Flux material  
(used in Refining Mg)

(It's a potassium fluoride &  
sodium hydroxide material)

The Flux is also used to  
extinguish Molten Metal.

Owners

Mike & Charles SLOVIC

Plant Manager: Greg Roush 216-214-7806

G. ORR called Greg R. Greg

said he didn't have owners number.

G. ORR asked if they hired a consultant yet.

G. Roush said NO that it would  
be the owners call to make.

G. ORR told him about the severe  
env. impact & that the USEPA would

be charging the company, so it

would be ~~good~~ <sup>in their</sup> best

interest to get their own  
consultant, working on the case.

A.S.A.P. Greg R. said he didn't

know the cause of the fire.

Greg said he didn't have numbers  
for owners; however he said

12/30/03

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Sarfield Allgas

the name for owner; however  
he said the name of the company  
where they would be treated  
was Magtech in Bellvue, Ohio  
Greg ORR 330.963.1189.

END of transcription

Reac sample collected over 8 hrs.

- CGI, FID/PID, DataRAM also collected

#1-#8 MCE File only

- #9 DataRAM only

#10-#11 MCE File

- #12-#13 Data RAM & MCE

#13 Upwind

12 source

11 South down

10 100 ft south of #11

9 200 ft south #11 [closest to building]

DataRAM - 8 hour runtime

Reac will ship samples via cargo  
to ERT in Edison New Jersey

20 12/30/03

Sarfie H Ali-

2000 Fire is still smoldering. REAC  
checks on samples.  
IR images evaluated from  
overflight.

1330 START off site.

Kelly Smith 12/30/03

12/31/03

21

### Garfield Alloys.

0740 START arrives at fire station No. 1.

Weather: overcast, temp ~ 32°F,

Wind chill ~ 20°F

Fire is still smoldering. All evacuations have been lifted.

REAC processing samples. Last

night: one dataram failed and one sample pump was

stolen (reac will supply data)

0830 REAC downloads dataram

summaries for START

1145 Talked w/ H+S guy for Garfield Alloys

sulfur dioxide 150 lb cylinders were found intact inside storage port.

of the site. START Butler on site

w/ OSC Kinkle and OSHA Reggie Brown

to survey the site and talk w/ site personnel. The counsel for the company

as well as some workers were

on site. OSHA OSC Reggie Brown

requested that the company

1) control the runoff from the scene from entering the storm sewers

Garfield Alloys 12/31/03

2) Provide security to the facility and 3) set the sulphur dioxide cylinders and any other cylinders off site. He expected that to be done by today. He will be on site tomorrow to make sure that happens. While on site a small fire erupted in the back of the facility & the fire department got back there.

OSC Brown informed the U.S. EPA OSCs that there will be a meeting at the OSHA with officials of Garfield Alloy and U.S. EPA & Ohio EPA. OSC Brown would like to have U.S. EPA at the meeting. OSC Kinnel requested that either OSC Dunn or OSC Furdk come to the meeting. He also requested the STANT supply a person. 1200 walked back to back of build to look at cylinders & take picture of them. The sulphur dioxide is intact & safe.

Garfield Alloy 12/31/03

1230 OSC Dunn & Fredle removed  
the trailer from the site to return  
to the Westlake office

The Fire Marshall was onsite with  
representatives from the Garfield  
Hb Fire Department

1350' Mike Morris of U.S. Chemical

Switch and Hb2 Board arrives w/  
two reps from Ohio Fire Marshall

Carson Investigators. So back to the  
Garfield Hb Fire Dept. to pack up  
equipment and leave for the day.

OK

~~12/31/03~~  
12/31/03



1-504 Sunfield Alloy  
 0910 meeting at OEPT w/ state  
 local representatives.  
 discuss that run-off material  
 had low pH and high lead  
 potentially caused by the run  
 off of potash

There is high ammonia w/ react  
 of magnesium ~~stress~~ and  
 fires. It is a chemical reaction  
 that occurs and is routine.

Ressie Brown - OEPT OSC  
 wants to have every person  
 & group mention what each  
 group is concerned with.

OEPT = asbestos = concerned  
 about any asbestos; even  
 less than  $< 1\%$  must be  
 considered contaminant ↓

<sup>OEPT</sup>  
 Jim Veris = confirmed that  
 building should be inspected.  
 One of the buildings that houses  
 on site was a demolition corp

OK — 12/11/10

Garfield Hwy 1-5-03

- that did asbestos remediation
- No real answer about the asbestos
- in the building from the demolition company owner.
- J. Venes <sup>EPA</sup> mentioned that any remediation that was done may have inspections that included asbestos. If any of that was done then Cleveland Air Div. may have been received

- Steve Tucker <sup>DSW</sup> said that the info that he received that the 1st round did not have values above water-criteria. He is concerned about water runoff. NEKSP said that they may have new data that suggests that copper data may be above acute levels.
- They are still concerned about ammonia, <sup>zinc</sup> pH and copper; <sup>suspended</sup> solids
- U.S. EPA concerned about the ammonia.

92 1-5-03

SM 1-5-04  
DSW wants the results

The company put in expandable plugs in the sewer. The water is contained.

Rich Connolly said there are no combined sewers in the immediate area.

95 + 97 orders were issued to the company because of discharge into Blue lagoon. The Co. now works with the NEORSD to discharge to their system.

The company presently has 2 lagoons; the easterly lagoon receives H<sub>2</sub>O from the storm sewer. The westerly lagoon receives the historical wastes.

DSW = asked NEORSD if they could discharge to the sewer. NEORSD said yes.

am 1/5/04

amz 1/5/04

DSW wants treatment system  
to be sent to a treatment system  
before discharge. The NEORSB  
is concerned about metals,  
ammonia;

Cell = no violation of fine  
water, does the fine water.  
run off sample collected by  
NEORSB cover the Cell  
violation run off.

NEORSB wants to see  
analytical results + a treatment  
system

PFI or approval of director  
permit to install

DSW + NEORSB want a PFI  
usually takes a week; but will  
get done.

Cuyahoga Fire Dept. concerned about  
the ammonia concentrations.

Reggie Brown said the company  
is monitoring for ammonia.

amz 1/5/04

Gartfield Alloy 1/5/04  
 The SHFD Chief, Colloz said  
 that Jim Ivin was monitoring  
 a large white plane that  
 came off of the site. Ivin  
 readings were 200 ppm in  
 plane 2 ppm outside  
 of plane.

Cuyahoga HHD w/ R. Romano  
 was ok with Magnesium Oxide  
 public health numbers. Reggie  
 wanted the CCHD to take lead  
 in the construction waste

1000 meeting w/ representatives  
 from Gartfield Alloy  
 Chris of Gartfield Alloy's state  
 Two security officers on site  
 at all times  
 1 controls the road + access; the  
 2nd patrols the fence line  
 - goes over plan the history  
 the map; the list of wastes +  
 materials on site.

OK 1/5/04

1/5/04 29  
end post 2  
Sanfield Alloy 15/04  
m-130 flux in at each  
fire post; class D extinguishers.

Spec. of Sanfield Alloy the flux  
10,500 at m-130 flux stayed  
in warehouse 3; brought in  
25,000 lbs of additional  
m-130 flux.

The NEORSO did an analysis of  
of the flux and found high  
levels of copper in mg/kg

The GA company does not have  
any add. info on the flux  
metal constituent.

GA retained A&T environmental

On 12/31/03 A&T responded to

GA request

1) beams at back

2) retention basin - pumped  
in holding tank; then  
transferred to Sanfield city  
waste water treatment.

3) Plug the sewers on site.

Presently the GA has collected  
on 1/5/04

Sarfield Alloy 1/5/04  
10,000 gallons. On Thursday after  
the water was collected begin  
then most water (6,500) collect  
on Sunday 1/4/04.

NEORSP = R. Corral, requires  
that the inflation devices be  
checked & routinely  
checked the run off + GA for  
no water has left site to send  
and to Mill Creek.

Ammonia is by product of magne  
reaction. The more moisture  
present the more likely to form  
ammonia it formed.

GA did ammonia testing from  
warehouse C; also did  
sampling from manufacturing  
and warehouse C. did grid  
sampling.

1/2/04 pulled 8 samples w/  
highest in container of waste  
C was Zypm. Those res-  
were reported to SHFD +  
OETA.

Anal 1/5/04

31  
Garfield Hwy 1/5/04  
yesterday due to rain, plumes  
of smoke was coming  
off the site.

GA is collecting ammonia air  
samples around the top of  
hill at the fence line  
near apartments at 20 yds  
a technical went into place  
to get 110 yds ammonia

OSC Brown told GA Chris that  
it would be likely that OSHA  
will request an ammonia  
monitor to provide continuous  
monitors

GA would like to monitor  
at the apartments, back  
by the Bannock's Demolition  
Co. + 3rd location

*possible  
reference*  
also will step up monitoring  
for ammonia when 1.  
rubble is moved 2. when  
it rains or snowed (or  
specific humidity level).

gbr 1/5/04



1/5/04 Garfield Alloy  
 OSC Brown requests that any  
 work that will require  
 an explosive or fire risk that  
 that work would require approval  
 from the GHFD. OEPA will not be  
 experts - Chief Calloway will review.  
 OSC Middle recommends that  
 infrared be used for hot spots

GA Chris talked w/ OSC Brown  
 at ~~1/12/04~~ 1/2/04 and proposed  
 in plan how will they approach  
 this clean up. They discussed  
 w/ Regyle about

- 1) Grab 4 samples of the  
 ash for KURAT analysis + pH  
 for analysis
- 2) Wipe samples of all building  
 materials to be disposed. Will  
 need to determine # of  
 samples

OEPA air = said that the  
 waste will be required =  
 asbestos expert must do a  
 (evaluation specialist) 1/5/04

1/5/04 Garfield Alloys

Visual inspection of the site building or other areas involved in the fire. If no bulk samples, then sample of debris should include asbestos sampling. Will need to survey the Bannister building; the matrix of roof materials.

Concern about the road access of the small companies that are behind the affected site. May need to reroute the road.

OSC Brown said that GA will need to talk to the adjacent owners so that GA knows all about the industry and that they are responsible for the environmental portion, but GA must get permission from the owners of the property to do any entry on the site; or damage.

OSC Brown feels that asbestos is an 1/5/04

1/5/04 Garfield Atty's  
will drive the clean-up.  
OEPA Jim Venes stated that  
if asbestos was found in the  
debris, then the magnesium  
ingots could be considered  
ACM.

GA Chris stated if ingots are  
found in the pile & are  
"cleaned" could be ~~appropriate~~  
appropriate & cleaned.

Chris states that refer to  
NEPA 430 reference material.  
If Chris for GA requested by  
OEPA to provide to Reggie and  
Chief & Colozza to provide  
documentation & proof that the  
ingots could be cleaned & safely  
OEPA Jim Venes will need  
to see the proof that they  
are "cleaned".

Reggie stated that all building  
material, if asbestos is found  
would be required to be  
considered ACM.

AK 1/5/04

Sartfield Allays 1/5/04

Reggie stated that the  
ingots ~~may~~ be able to be  
cleaned but will need to  
be approved by Chief Calozza  
& Reggie

Randall from the site should be  
in 10 ppm ammonia.

Steve Tucker from OERT/PSW  
wants the company know that  
the temp; hardness; pH + zinc  
ammonia, copper should be  
collected if there is a problem  
or an accidental release.

Reggie requested the plan be  
redone and include an  
ammonia sampling plan. Reggie  
will issue a Notice of  
Violation letter to tell them  
of problem at cell

site number

210

4843

210 214 7804 cell for Chris

(537) 790 0760 cell 1st

cherz@jrtieallays.com enviro  
Cle@jrtieallays.com 1/5/04

GA will have a plan to Reggie  
by 1/15/04

Attendance of meeting and the  
Sanfield Alloys plan to  
cleanup (draft) are to  
be bound in site files —



Aurora 1/5/04

Attachment B

Photographic Log



**Photograph No.:** 1

**TDD No.:** S05-0312-007

**Date:** Not known

**Location:** Garfield Alloys Fire site

**Subject:** Garfield Alloys prior to fire



**Photograph No.:** 2

**TDD No.:** S05-0312-007

**Date:** December 29, 2003

**Location:** Garfield Alloys Fire site

**Subject:** Fire at Garfield Alloys facility in Garfield Heights, Ohio



**Photograph No.: 3**

**TDD No.:** S05-0310-010

**Date:** December 30, 2003

**Location:** Garfield Alloys Fire site

**Subject:** Fire at Garfield Alloys after approximately 12 hours.



**Photograph No.: 4**

**TDD No.:** S05-0312-007

**Date:** December 31, 2003

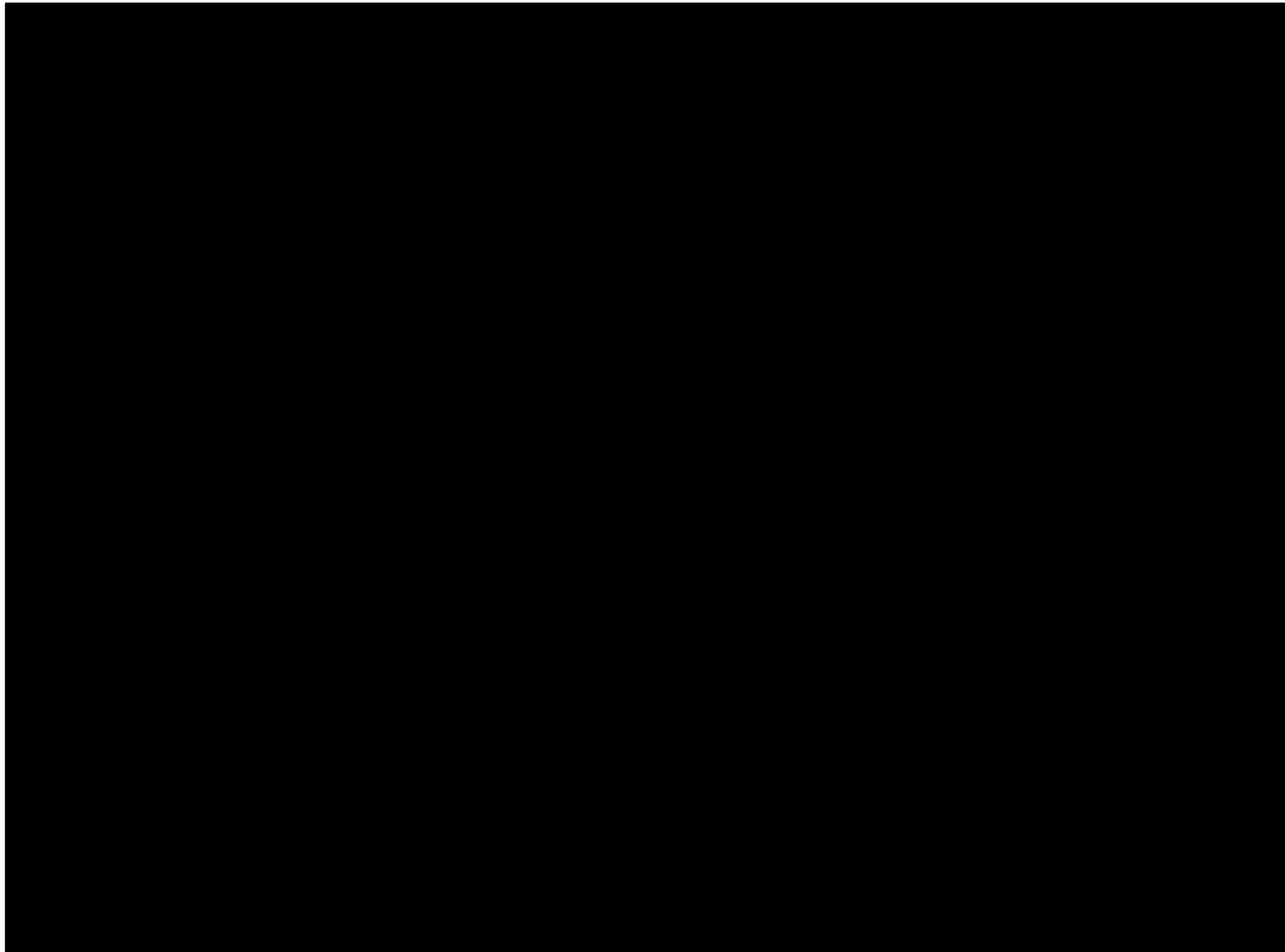
**Location:** Garfield Alloys Fire site

**Subject:** Burn area covered with flux used to smolder the fire



Attachment C

Air Monitoring and Sampling Locations Map and REAC DataRAM Results



Attachment D

START Air Monitoring Log

# **Table 2** **Garfield Alloy Fire** **Air Monitoring Log**

Date: 12/29/03

Sample #/ Location	Time	O2	H2S	FID	PID	LEL	Data RAM	Sampler Name
1.								
2.								
3.								
4.								
5.								
6.	2230 2314	20.3%	0.0	2.63 ppb	0.0ppm 0.0ppm - 0.0ppm			Busher wolfe
7.								
8.								

Sample Locations:

12/29/03 unmarked background

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. Sample location #8:

Equipment used:

CGI: PHD lite - START

FID: TVA 1000 - START

PID: PGM50-SP U.S. EPA MultiRAE - gross fire

DATA RAM: \_\_\_\_\_

## Table 2 Garfield Alloy Fire Air Monitoring Log

Date: 12/29/03

Prayer

Sample #/ Location	Time	O2	H2S	FID	PID	LEL	<del>Data</del> RAM	Sampler Name
1.								
2.								
3.								
4.								
5.								
6.	2323			4.12 3.12x			0.0cl 0.0 and 0.0 NH <sub>3</sub>	Bush wife
7.								
8.								

Sample Locations:

\* collected at 722ch

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. Sample location #8:

Equipment used:

CGI: \_\_\_\_\_

PID: \_\_\_\_\_

FID: START TVA 1000

DATA RAM: \_\_\_\_\_

## Table 2

### Garfield Alloy Fire

### Air Monitoring Log

Date: 12/30/03

Sample #/ Location	Time	O2	H2S	FID	PID	LEL	Data RAM	Sampler Name
1.								
2.								
3.	0136	20.3%	0.0	2.50	—			
4.	0557	<del>20.3%</del> 20.3%	<del>0.0</del> 0.0*		0.0*	<del>0.0</del> 0.0*	1.64 ppm 0.45 ppm 2.6 ppm	
5.								
6.								
7.	655	<del>20.3%</del> 20.3%	<del>0.0</del> 0.0*		0.0*	<del>0.0</del> 0.0*	0.0	
8.								

Sample Locations:

\* MultiRAE

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. Sample location #8:

Equipment used:

CGI: PHD lite

PID: MultiRAE

FID:

DATA RAM:

# Garfield Alloy Fire Air Monitoring Log

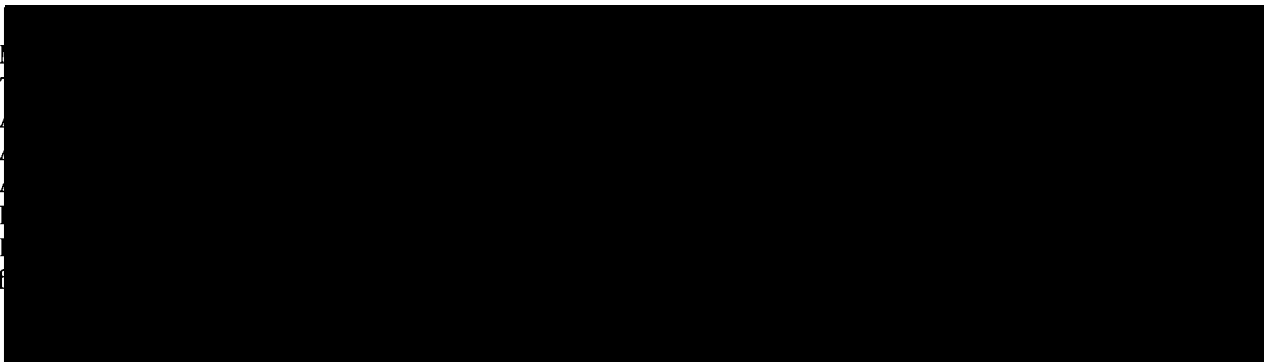
*particulates*

Date: 12-30-03

Sample #/ Location	Time	O2 (%)	H2S (ppm)	FID (units)	PID (units)	LEL (%)	Data RAM	Sampler Name
1.	1126	20.6	0	0.41	0	0	Real Time 0.012 TWA 0.010	TSM
2.	1205	20.2	0	1.40	0	0	0.008 0.008	TSM
3.	1215	20.3	0	2.70	0	0	0.016 0.004	TSM
4.	1230	20.3	0	2.18	0	0	0.010 0.010	TSM
5.	1245	20.3	0	2.05	0	0	0.012 0.012	TSM
6.	1315	20.6	0	2.24	0.06	0	0.037 0.037	TSM
7.	1330	20.7	0	2.51	0	0	0.005 0.033	TSM
8.	1350	20.6	0	2.70	0	0	0.041 0.031	TSM

Sample Locations:

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. At the intersection of



Equipment used:

CGI: \_\_\_\_\_

PID: \_\_\_\_\_

FID: \_\_\_\_\_

DATA RAM: \_\_\_\_\_

*Frank J. Beahm - Veston Solutions, Inc.*

12/30/03

# **Table 2** **Garfield Alloy Fire** **Air Monitoring Log**

Date:

Sample #/ Location	Time	O2	H2S	FID	PID	LEL	Data RAM	Sampler Name
1.	1400	20.2	0	X	0	0	0.018 0.033	TSM
(2.)	1655	20.2	0	X	0	0	0.018 0.033	TSM
3.	1853	20.2	0	X	0	0	0.020 0.031	TSM
4.	1850	20.1	0	X	0	0	0.028 0.043	TSM
5.	1845 <del>1845</del>	20.1	0	X	0	0	0.041 0.041	TSM
6.	1835	19.4	0	X	0	0	0.014 0.038	TSM
7.	1815	19.5	0	X	0	0	0.021 0.033	TSM
8.	1820	19.6	0	X	0	0	0.022 0.038	TSM

Sample Locations:

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. Sample location #8:



Equipment used:

CGI: \_\_\_\_\_

PID: \_\_\_\_\_

FID: \_\_\_\_\_

DATA RAM: \_\_\_\_\_

*Frank D. Buckley - Western Solutions, Inc.*



# Garfield Alloy Fire Air Monitoring Log

Date: 12-30-03

Sample #/ Location	Time	O2	H2S	FID	PID	LEL	Data RAM	Sampler Name
1.	6:00	20.6	0	X	0.1	0	Real Time 0.002 TWA 0.040	TSM
2.	6:15	X	X	X	X	X	X	No Access
3.	6:30	20.5	0	X	0	0	0.007 0.040	TSM
4.	6:50	20.8	0	X	0	0	0.039 0.062	TSM
5.	<del>7:50</del> 7:10	20.2	0	X	0	0	0.006 0.069	TSM
6.	7:30	20.2	0	X	0	0	0.003 0.056	TSM
7.	7:40	X	X	X	X	X	X	No Access
8.	7:50	20.3	0	X	0	0	0.007 0.047	TSM

Sample Locations:

1. Sample location #1:
2. Sample location #2:
3. Sample location #3:
4. Sample location #4:
5. Sample location #5:
6. Sample location #6:
7. Sample location #7:
8. At the intersection of

Equipment used:

CGI: \_\_\_\_\_

PID: \_\_\_\_\_

FID: \_\_\_\_\_

DATA RAM: \_\_\_\_\_

Mark L. Bradley - Western Station, Inc.

Location / DR  
10 - 100 ft. downwind  
of site

[REDACTED]

10 - 100 ft. downwind  
Sample

[REDACTED]

100 gals

11 - 100 ft. downwind  
Sample

[REDACTED]

350 gals

12 - 100 ft. downwind  
Sample / DR

[REDACTED]

13 - 100 ft. downwind  
Sample / DR

[REDACTED]

Sampling  
CO 1, LEL 0, VOL 0,  
H<sub>2</sub>S 0, O<sub>2</sub> 19.9

0.403 473

H<sub>2</sub>S 0, O<sub>2</sub> 19.9  
CO 0, LEL 0, VOL 0  
0.343

CO 0, LEL 0, VOL 0  
H<sub>2</sub>S 0, O<sub>2</sub> 19.9

0.459

0.054

CO 0, LEL 0, VOL 0  
H<sub>2</sub>S 0, O<sub>2</sub> 20.1

0.442 0.297

CO 0, LEL 0, VOL 0  
H<sub>2</sub>S 0, O<sub>2</sub> 19.9  
0.016

Instrument  
4-HiRAE

PDR

11/30/03

Frank L. DeLong

Western Solutions, Inc.

Attachment E

START Analytical Data Results



## Laboratory Analytical Report

### **TetraTech EMI, Inc.**

6801 Engle Road

Suite G

Cleveland, OH 44130

Attention:  
Kelly Smith

### **Project Identification**

Garfield Alloys

### **Purchase Order:**

**EA Group**  
**Order Number**  
0312-00283

A handwritten signature in cursive script, appearing to read "Donald R. Richner".

Donald R. Richner, CIH  
Laboratory Manager

December 30, 2003

**Project Summary**

The following analytical report contains the results as requested for samples submitted to EA Group. The results included in this report have been reviewed for compliance with the analytical methods indicated in this report. All data have been found to be compliant with accepted laboratory protocol. Exceptions, if any, are noted below. Analytes appearing in bold type were analyzed at a subcontract facility. EA Group is VAP, AIHA and ELLAP accredited. For industrial hygiene reports, air and/or surface concentrations results are based upon field sampling information provided by the client. Unless otherwise noted the following apply: Sample condition was acceptable upon receipt and Industrial hygiene results will not be blank corrected.

**Data Interpretation**

For assistance with report interpretation or questions regarding regulatory limits, please contact Client Services at 440-951-3514 or customerservice@eagroup-ohio.com.

**Sample Summary**

Sample Receive Date: 12/30/2003

<u>EAG</u> <u>Sample Identification</u>	<u>Client</u> <u>Sample Identification</u>	<u>EAG</u> <u>Sample Identification</u>	<u>Client</u> <u>Sample Identification</u>
031200283 - 001	MT-01	031200283 - 002	MT-02
031200283 - 003	MT-03	031200283 - 004	MT-04
031200283 - 005	VO-01		

**Quality Control Narrative**

A GC/MS scan of the sample for volatile organics failed to identify any compounds above our quantitation limit.

\*\*\*\*\*

Reproduction of this report is prohibited except in its entirety. Unless noted, soil, sludge, and sediment results are reported on dry weight basis. The "Sample Reporting Limit" is based on the method used for analysis and does not refer to any regulatory limit. These results relate only to the items tested.



EAG Workorder: 0312-00283

Client Project: Garfield Alloys

EAG ID: 0312-00283-1 Client ID: MT-01 Sampled: 12/29/2003 Received: 12/30/2003

<u>Parameter</u>	<u>Result</u>	Sample Reporting <u>Limit</u>	<u>Units</u>	<u>Prep</u> <u>Date</u>	<u>Analysis</u> <u>Date</u>	<u>Analyst</u>
Magnesium as MgO, Air: NIOSH 7300	0.26	0.14	mg/m3	12/30/2003	12/30/2003	CMB

EAG ID: 0312-00283-2 Client ID: MT-02 Sampled: 12/29/2003 Received: 12/30/2003

<u>Parameter</u>	<u>Result</u>	Sample Reporting <u>Limit</u>	<u>Units</u>	<u>Prep</u> <u>Date</u>	<u>Analysis</u> <u>Date</u>	<u>Analyst</u>
Magnesium as MgO, Air: NIOSH 7300	0.22	0.13	mg/m3	12/30/2003	12/30/2003	CMB

EAG ID: 0312-00283-3 Client ID: MT-03 Sampled: 12/29/2003 Received: 12/30/2003

<u>Parameter</u>	<u>Result</u>	Sample Reporting <u>Limit</u>	<u>Units</u>	<u>Prep</u> <u>Date</u>	<u>Analysis</u> <u>Date</u>	<u>Analyst</u>
Magnesium as MgO, Air: NIOSH 7300	0.84	0.13	mg/m3	12/30/2003	12/30/2003	CMB

EAG ID: 0312-00283-4 Client ID: MT-04 Sampled: 12/29/2003 Received: 12/30/2003

<u>Parameter</u>	<u>Result</u>	Sample Reporting <u>Limit</u>	<u>Units</u>	<u>Prep</u> <u>Date</u>	<u>Analysis</u> <u>Date</u>	<u>Analyst</u>
Magnesium as MgO, Air: NIOSH 7300	<0.12	0.12	mg/m3	12/30/2003	12/30/2003	CMB



EAG Workorder: 0312-00283

Matrix: Tube

Date Sampled: 12/29/2003

EAG ID: 0312-00283-005

QC Batch / Analyst: 045701 / DFM

Date Received: 12/30/2003

Client ID: VC-01

Client Project: Garfield Alloys

<u>Parameter</u>	<u>Result</u>	<u>Reporting Limit</u>	<u>Units</u>	<u>Date Analyzed</u>
GC/MS Semivolatiles Scan	<0.044	0.044	ppm	12/30/2003



EAG Workorder: 0312-00283

Matrix: Tube

Date Sampled: 12/29/2003

EAG ID: 0312-00283-005

QC Batch / Analyst: 045702/DLZ

Date Received: 12/30/2003

Client ID: VO-01

Client Project: Garfield Alloys

---

<u>Parameter</u>	<u>Result</u>	<u>Reporting Limit</u>	<u>Units</u>	<u>Date Analyzed</u>
Organics in Air: OSHA 7				12/30/2003
Total Hydrocarbons as Toluene	<0.044	0.044	ppm	12/30/2003
Special Media Desorption	Complete			12/30/2003



CHAIN OF CUSTODY RECORD

PROJ. NO.		PROJECT NAME				NO. OF CONTAINERS	Activity Code:												
		Garfield Alloys					<div style="display: flex; justify-content: space-between;"> <div>Analyte:</div> <div> <div>Magnesium Metal</div> <div>Volatile Organics</div> </div> </div>												
SAMPLERS: (Print Name and Sign)																			
<div> <div>John A. Boston</div> <div>Stephen Wolfe</div> </div>																			
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION	TAG NUMBERS													
MT-01	12/24/03	2230		X	Between E110 + E111	1-assy	X									Volume = 32.525 Liters			
MT-02	12/29/03	2231		X	Between E110 + E111	1-assy	X									Volume = 32.67 Liters			
VC-01	12/24/03	2207		X	Between E110 + E111	1-tube	X	X								Volume = 12.1 Liters			
MT-03	12/30/03	0557		X	Broadway - Forest View	1-assy	X									Volume = 32.55 Liters			
MT-04	12/24/03	0055		X	Edgepark + Longbe	1-assy	X									Volume = 33.92 Liters			
Relinquished by: (Signature)		Date / Time		Received by: (Signature)		Ship To: Fox & Sons Inc. 816-475-8389 Phone To: 1500 W. 14th Street 7th - 7th Floor Chicago, IL 60604 212-247-7847													
Relinquished by: (Signature)		Date / Time		Received by: (Signature)															
Relinquished by: (Signature)		Date / Time		Received for Laboratory by: (Signature)												Date / Time			
										Chain of Custody Seal Numbers									

Distribution: White - Accompanies Shipment; Pink - Coordinator Field Files; Yellow - Laboratory File

Attachment F

ASPECT Report

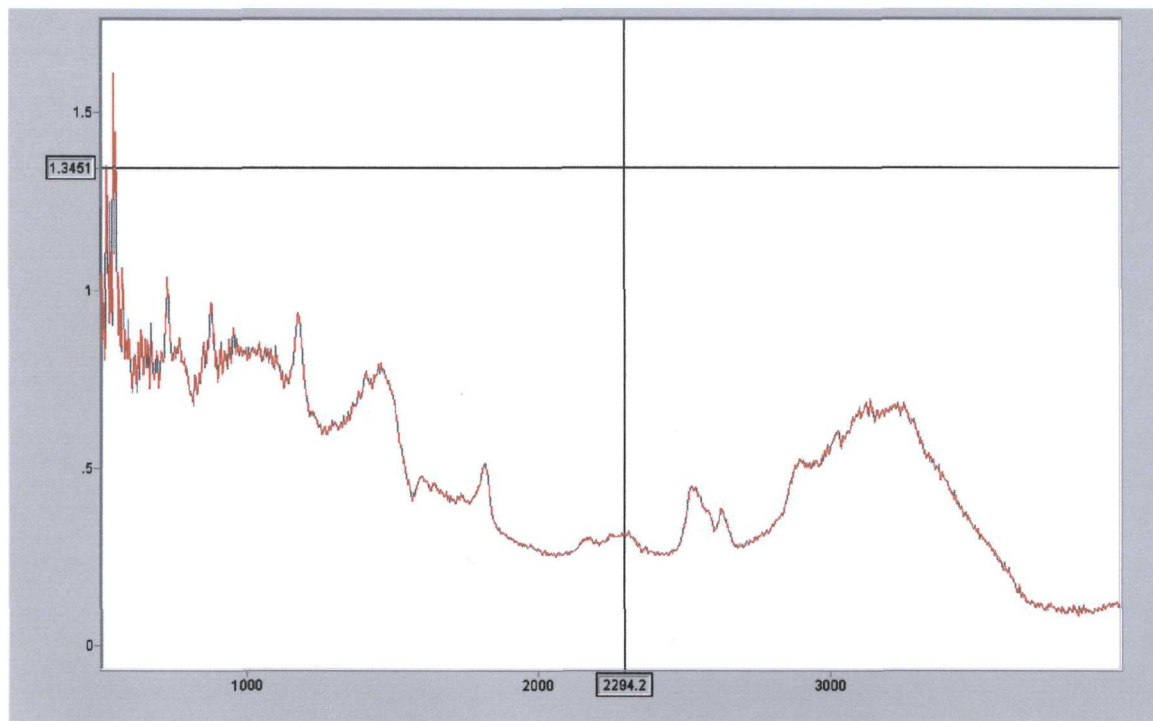
## Preliminary Interpretation of IR Spectra from the Ohio Magnesium Fire.

Data Collected on 30 December 2003.

The ASPECT system was used to image the magnesium fire near Garfield Heights, Ohio. At times this fire was spectacular with intense heat due to the burning magnesium. Magnesium is a very active metal and once ignited burns at extremely high temperature. One of the primary combustion products of burning magnesium is magnesium oxide. This material is a white, powdery solid with a melting point of 2800 C.

ASPECT is an airborne hyper/multi-spectral Infrared (IR) system used to image and identify chemical plumes. Typically, the system is used to detect gaseous emissions. ASPECT uses a high speed spectrometer (Bomem MR-254AB) running at 80 hertz. A spectral resolution of 16 wavenumbers is used at this scan speed. The spectrometer collects a scan at a ground resolution of about 0.75 meters. The spectrometer is used to provide a hyper-spectral signature of the gas. Based on this signature, a library match can be made with unknown identification. The other principle sensor used is a Raytheon RS-800 Line Scanner. This is a multi-spectral (14 band) imager which will produce an image of approximately 1200 meters based on the normal operational altitude of the aircraft. The Line Scanner uses a collection of cold optical filters to provide spectral discrimination of the collected data.

At approximately 1430 GMT Region 7 contacted Region 5 concerning the magnesium fire near Garfield Heights. The ASPECT program manager indicated that ASPECT had potential to image the plume emanating from the fire. This was based on a spectra of MgO pulled from the Galactic Data Base and is given in figure 1.



An examination of this spectra shows several features in both the 8 to 12 micron (800 to 1400 wavenumber) region and the 3 to 5 micron (2200 to 3500 wavenumber) region. The absorbance of this compound tends to be high with values approaching 1 absorbance unit in the 8 to 12 micron region. Several peaks stand out including peaks at 880, 1176, 1460, and 1816 wavenumber, respectively. The 3 to 5 region shows a peak structure at 2531 wavenumber.

Figure 2 shows a single beam scan taken over the residual fire. Figure 3 shows an overlay of the two scans. Note that the scan taken over the fire has not been spectrally subtracted due to the high energy content of the signal. A comparison of the two spectra shows little peak correlation in the 900 to 1200 wavenumber region. This is not unexpected due to the large energy content of the sample. A peak of interest occurs at 1816 wavenumber. This is a portion of the longwave region that is normally obstructed by water vapor and carbon dioxide. In this case a reasonable match is noted with similar band structure. This is likely due to two factors. Since the energy level of the signal is so high, data normally absent due to water vapor/carbon dioxide interference have sufficient energy to be discriminated from the interferent background. Second, since this is on the lower portion of the normal blackbody curve of the detector envelop, a signal present in this region indicates the presence of a compound. Further examination of the spectra show an elevation of the general trend of the blackbody between 850 and 900 wavenumbers. This may be due to the compound influencing the signal due to the 876 wavenumber peak.

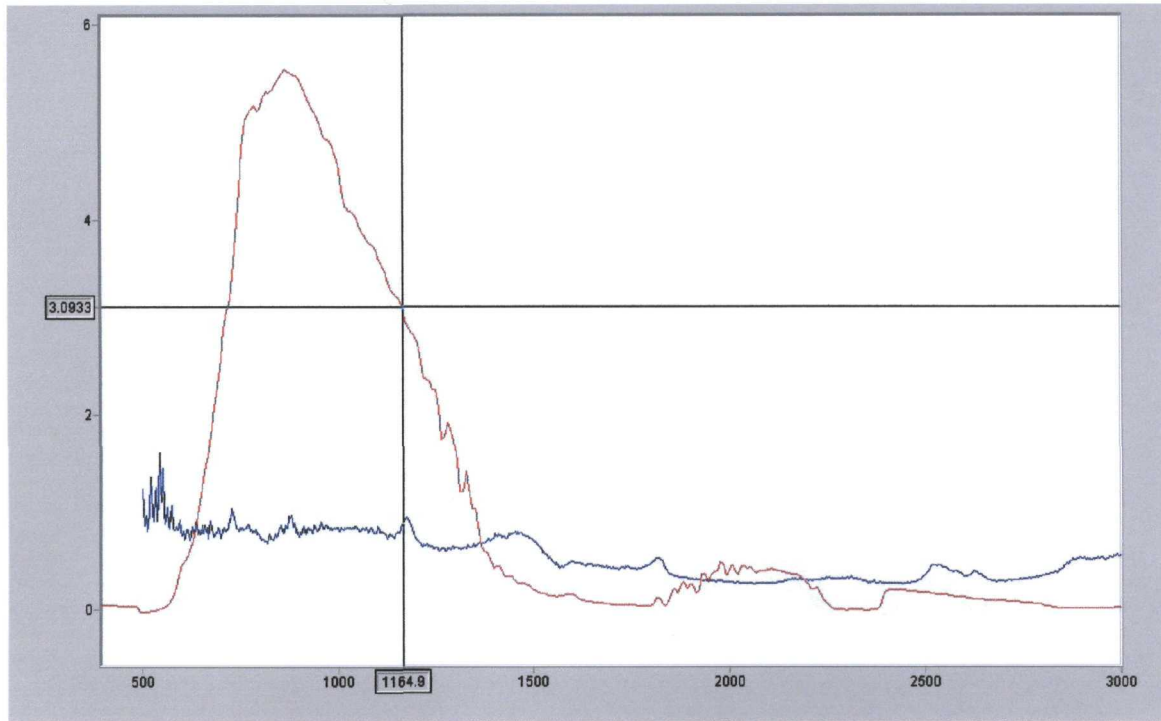


Figure 4 shows a on overlay of a similar scan collected on a subsequent base of the aircraft. A peak analysis show agreement with the previous aircraft pass with peaks at 876, and 1816 being similar. This scan also shows a potential indication at 1603 wavenumbers.

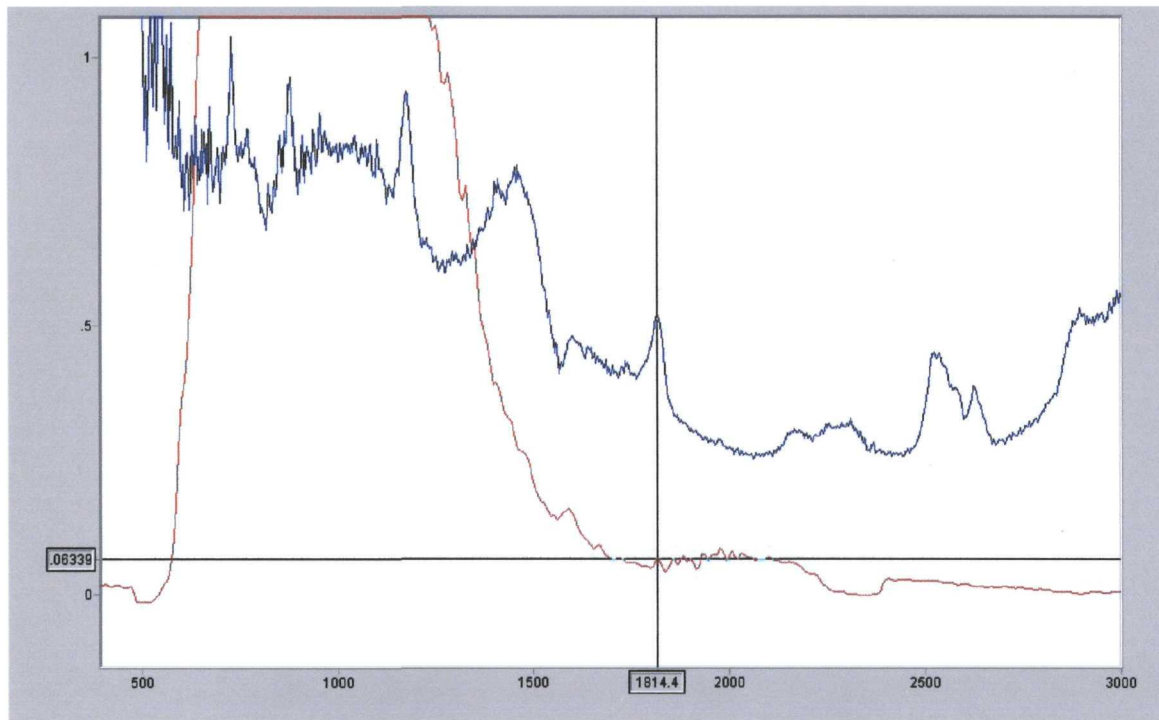
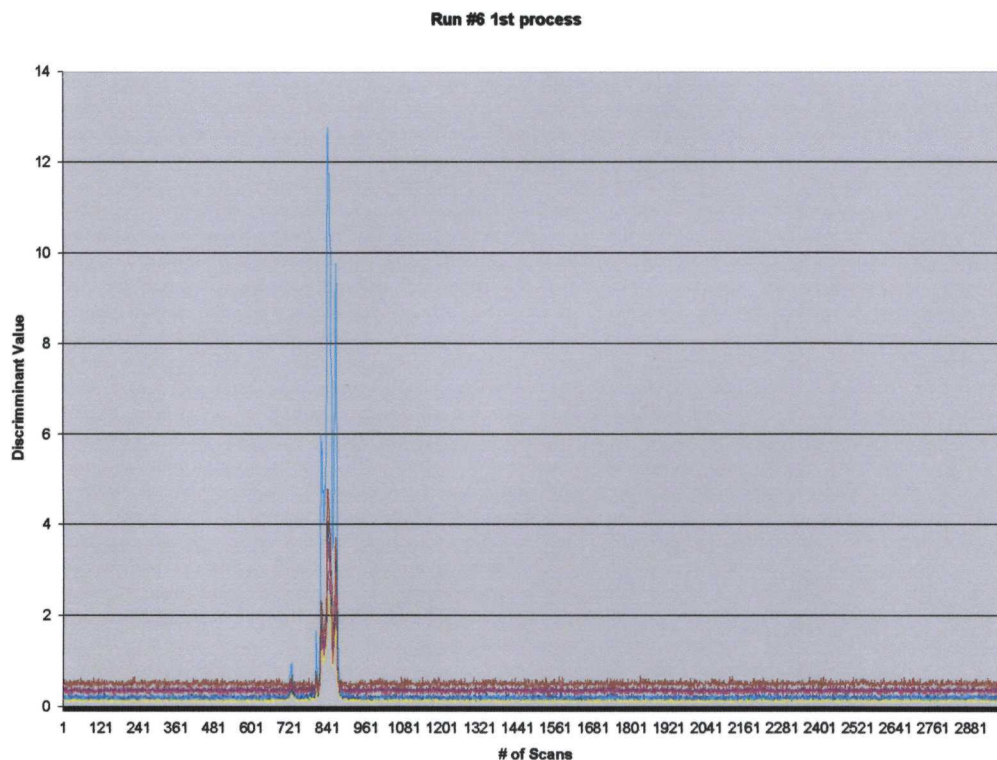




Figure 5 shows the results of a discrimination analysis using digital filtering to identify spectra showing a significant difference from a background training set. An examination of this figure clearly indicates repetitive regions of similar spectra. A portion of this discrimination is caused by the high energy radiance coming from the fire. The remainder of the discrimination is due to spectral feature.



Assigning a concentration of MgO based on this spectra is difficult due to high energy of the situation. MgO is formed due to the combustion of Mg and is most likely emitted initially as a vapor (boiling point of 3600 C). It is probable that the spectral features observed are in fact due to IR absorptive-mode features of the vapor. MgO is converted to a solid rapidly upon cooling and subsequently loses the absorptive-mode character. The full data set will be required to extract a subtracted spectra but based on the high absorbance of this material, a concentration of less than 250 ppm-meter is emanating from the fire.

Figure 6 shows an IR image collected using the IR line Scanner. No detectable plume was present. The Line Scanner is a multi-spectral imager with 14 bands in both the 8 to 12 and 3 to 5 micron region. The figure 6 was generated by using a bands in the 9, 10, and 12 micron regions to provide a RGB display. The only significant feature is the elevated temperature areas due to the fire. A total of 8 passes were made with the aircraft with similar results for each pass. In summary, a plume was not detected leaving the heated areas.

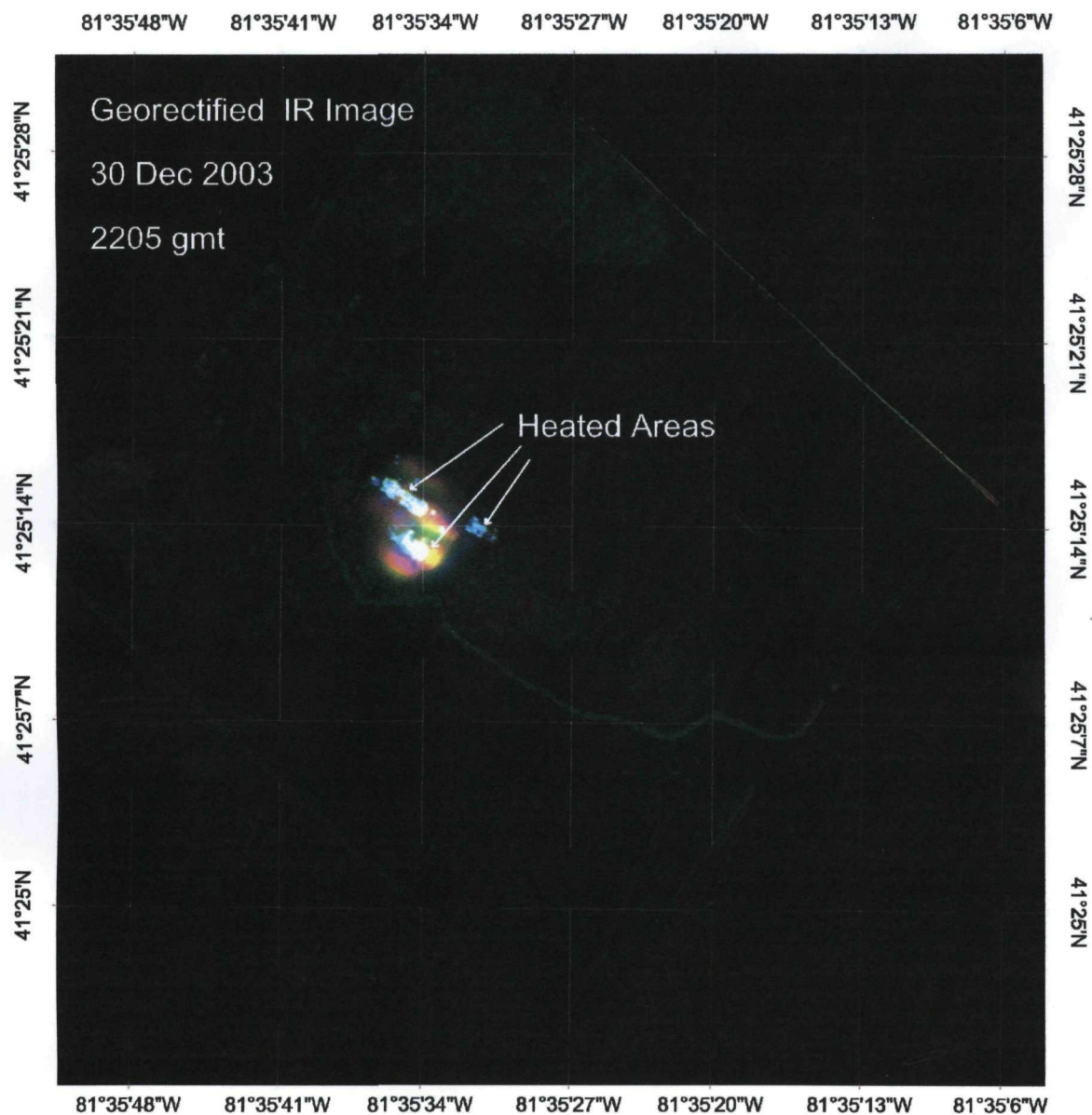
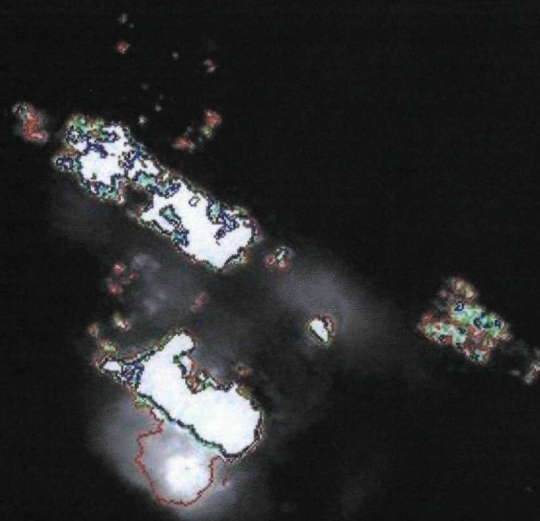


Figure 7 shows a base thermal classification of the fire. The RS-800 IR line scanner is a very accurate radiometric imager. Two calibration sources are used to set the dynamic range of the system. The contours shown are derived by using the lowest calibration temperature of 8.3 C and the highest being 30 C. All data falling between these temperatures are classified. Temperatures above 30 C are depicted as pure white.

## Temp. Contour Levels



Red = 10 - 15 degrees C

Green = 15 - 20 degrees C

Blue = 20 - 25 degrees C

Sienna = 25 - 30 degrees C

Pure White Areas Are Hotter than 30degrees C



Attachment G

SUMMA Canister Analytical Results

ANALYTICAL REPORT


Prepared by  
Lockheed Martin Technology Services Group

*Ohio Magnesium Fire Emergency Response*  
Cleveland, OH

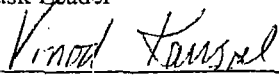
January 2004

EPA Work Assignment No. 82-001  
Lockheed Martin Work Order No. R1A82001  
EPA Contract No. 68-C99-223

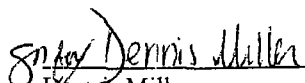
Submitted to  
Raj Singhvi  
EPA-ERTC

  
Dennis Miller  
Task Leader  
1-21-04  
Date

Analysis by:  
REAC

  
Vinod Kansal  
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1/21/04  
Date

Prepared by:  
Mark Bernick

  
Dennis Miller  
Program Manager  
1-21-04  
Date

Reviewed by:  
Joseph Soroka

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Appendices will be furnished on request.

## Introduction

REAC personnel in response to WA 82-001, provided analytical support for environmental samples collected from the Ohio Magnesium Fire Emergency Response, located in Cleveland, Ohio as described in the following table. The support also included QA/QC, data review, and preparation of an analytical report containing a summary of the analytical methods, the results, and the QA/QC results.

The samples were treated with procedures consistent with those specified in SOP #1008.

COC #	# of Samples	Sampling Date	Date Received	Matrix	Analysis	Laboratory	Data Package
5-69741	2	12/30/03	1/6/04	Air	TO-14	REAC	N006

## Case Narrative

The data in this report have been validated to two significant figures. Any other representation of the data is the responsibility of the user. No results less than 25 percent of the MDL were reported. Several target compounds were manually integrated during the calibrations; the data are not affected.

### VOC in Air Package N006

In the initial calibration on 12/2/03 the percent relative standard deviation exceeded the QC limits for vinyl chloride (26%). This compound was not detected in the associated samples; the data are not affected.

In the continuing calibration on 1/6/04 the percent difference exceeded the QC limits for vinyl chloride (27%). This compound was not detected in the associated samples; the data are not affected.

## Summary of Abbreviations

AA	Atomic Absorption
B	The analyte was found in the blank
BFB	Bromofluorobenzene
C	Centigrade
cont.	Continued
D	(Surrogate Table) this value is from a diluted sample and was not calculated (Result Table) this result was obtained from a diluted sample
Dioxin	denotes Polychlorinated Dibenzo-p-dioxins and Polychlorinated Dibenzofurans and/or PCDD and PCDF
CLP	Contract Laboratory Protocol
COC	Chain of Custody
CONC	Concentration
CRDL	Contract Required Detection Limit
CRQL	Contract Required Quantitation Limit
DFTPP	Decafluorotriphenylphosphine
DL	Detection Limit
E	The value is greater than the highest linear standard and is estimated
EMPC	Estimated maximum possible concentration
ICAP	Inductively Coupled Argon Plasma
ISTD	Internal Standard
J	The value is below the method detection limit and is estimated
LCS	Laboratory Control Sample
LCSD	Laboratory Control Sample Duplicate
MDL	Method Detection Limit
MI	Matrix Interference
MS (BS)	Matrix Spike (Blank Spike)
MSD (BSD)	Matrix Spike Duplicate (Blank Spike Duplicate)
MW	Molecular Weight
NA	either Not Applicable or Not Available
NC	Not Calculated
NR	Not Requested
NS	Not Spiked
% D	Percent Difference
% REC	Percent Recovery
PAL	Performance Acceptance Limit
PPB	Parts per billion
PPBV	Parts per billion by volume
PPMV	Parts per million by volume
PQL	Practical Quantitation Limit
QA/QC	Quality Assurance/Quality Control
QL	Quantitation Limit
RPD	Relative Percent Difference
RSD	Relative Standard Deviation
SIM	Selected Ion Monitoring
TIC	Tentatively Identified Compound
TCLP	Toxicity Characteristic Leaching Procedure
U	Denotes not detected
W	Weathered analyte; the results should be regarded as estimated
m <sup>3</sup>	cubic meter      kg      kilogram      µg      microgram
L	liter      g      gram      pg      picogram
mL	milliliter      mg      milligram      ng      nanogram
µL	microliter
*	denotes a value that exceeds the acceptable QC limit

Abbreviations that are specific to a particular table are explained in footnotes on that table

Revision 7/16/03

### Analytical Procedure for VOC in Air

The samples were analyzed with procedures consistent with those described in REAC SOP# 1814 and detailed in the SUMMA canisters analytical reports (Appendix A). The VOC results are listed in Table 1.1; the tentatively identified compounds (TICs) are listed in Table 1.2.

Table 1.1 Results of the Analysis for VOC in Air  
Ohio Magnesium Fire Emergency Response, WA # 82-001

Sample # :	Method Blank	REAC4348	REAC4347
Location :	040106-1	Trip Blank	14200 Broadway Ave
Date Sampled :	N/A	12/30/03	12/30/03
Date Analyzed :	01/06/04	01/06/04	01/06/04

Compound Name	Conc. ppbv	MDL ppbv	Conc. ppbv	MDL ppbv	Conc. ppbv	MDL ppbv
Chloromethane	U	4	U	4	U	4
Vinyl Chloride	U	4	U	4	U	4
Chloroethane	U	4	U	4	U	4
Trichlorofluoromethane	U	4	U	4	U	4
1,1-Dichloroethene	U	4	U	4	U	4
Methylene Chloride	U	4	U	4	U	4
trans-1,2-Dichloroethylene	U	4	U	4	U	4
1,1-Dichloroethane	U	4	U	4	U	4
cis-1,2-Dichloroethene	U	4	U	4	U	4
Trichloromethane	U	4	U	4	U	4
1,1,1-Trichloroethane	U	4	U	4	U	4
Carbon Tetrachloride	U	4	U	4	U	4
1,2-Dichloroethane	U	4	U	4	U	4
Benzene	U	4	U	4	U	4
Trichloroethylene	U	4	U	4	U	4
Bromodichloromethane	U	4	U	4	U	4
Dibromomethane	U	4	U	4	U	4
Toluene	U	4	U	4	U	4
1,1,2-Trichloroethane	U	4	U	4	U	4
Tetrachloroethylene	U	4	U	4	U	4
Ethylbenzene	U	4	U	4	U	4
m & p-Xylenes	U	4	U	4	U	4
o-Xylene	U	4	U	4	U	4
Styrene	U	4	U	4	U	4
1,1,2,2-Tetrachloroethane	U	4	U	4	U	4
1,3,5-Trimethylbenzene	U	4	U	4	U	4
1,2,4-Trimethylbenzene	U	4	U	4	U	4

Table 1.2 Results of the TIC Analysis for VOC in Air  
Ohio Magnesium Fire Emergency Response, WA # 82-001

Sample #	Compound
Method Blank 040106-1	No TICs Found
REAC4348	No TICs Found



Table 1.2 (cont.) Results of the TIC Analysis for VOC in Air  
Ohio Magnesium Fire Emergency Response, WA # 82-001

Sample: 15441

Conc. Factor: 2.0

Compound Name	Retention Time	Concentration (ppbv)*
unknown	2.983	4
acetaldehyde	4.161	9
acetone	7.341	14
butanal	11.094	4
unknown	11.285	4

\*Estimated Concentration (Response Factor = 1.0)

## QA/QC for VOC in Air

### Results of the Internal Standard Areas for VOC in Air

The internal standard areas (for bromochloromethane, 1,4-dichlorobenzene and chlorobenzene-d5) are listed in Table A6. All 18 values were within the acceptable QC limits.

### Results of the Matrix Spike Matrix Spike Duplicate Analysis for VOC in Air

Sample REAC4347 was chosen for the matrix spike matrix spike duplicate (MS/MSD) analysis. The percent recoveries, listed in Table A5, ranged from 41 to 113. The relative percent differences (RPDs), also listed in Table A5, ranged from zero to two. No QC limits have been established for the above parameters..

### Results of the Duplicate Sample Analysis for VOC in Air

Sample REAC 4347 was chosen for the duplicate sample analysis (Table 2.1). RPDs are not reported because no target compounds were detected above their respective MDLs.

Table 2.1 Results of the Duplicate Analysis for VOC in Air  
Ohio Magnesium Fire Emergency Response, WA # 82-001

Sample # :	REAC4347	REAC4347 Dup
Location :	14200 Broadway Ave	14200 Broadway Ave
Date Sampled :	12/30/03	12/30/03
Date Analyzed :	01/06/04	01/06/04

Compound Name	Conc. ppbv	MDL ppbv	Conc. ppbv	MDL ppbv	RPD	QC Limit RPD *
Chloromethane	U	4	U	4	NC	20
Vinyl Chloride	U	4	U	4	NC	20
Chloroethane	U	4	U	4	NC	20
Trichlorofluoromethane	U	4	U	4	NC	20
1,1-Dichloroethene	U	4	U	4	NC	20
Methylene Chloride	U	4	U	4	NC	20
trans-1,2-Dichloroethylene	U	4	U	4	NC	20
1,1-Dichloroethane	U	4	U	4	NC	20
cis-1,2-Dichloroethene	U	4	U	4	NC	20
Trichloromethane	U	4	U	4	NC	20
1,1,1-Trichloroethane	U	4	U	4	NC	20
Carbon Tetrachloride	U	4	U	4	NC	20
1,2-Dichloroethane	U	4	U	4	NC	20
Benzene	U	4	U	4	NC	20
Trichloroethylene	U	4	U	4	NC	20
Bromodichloromethane	U	4	U	4	NC	20
Dibromomethane	U	4	U	4	NC	20
Toluene	U	4	U	4	NC	20
1,1,2-Trichloroethane	U	4	U	4	NC	20
Tetrachloroethylene	U	4	U	4	NC	20
Ethylbenzene	U	4	U	4	NC	20
m & p-Xylenes	U	4	U	4	NC	20
o-Xylene	U	4	U	4	NC	20
Styrene	U	4	U	4	NC	20
1,1,2,2-Tetrachloroethane	U	4	U	4	NC	20
1,3,5-Trimethylbenzene	U	4	U	4	NC	20
1,2,4-Trimethylbenzene	U	4	U	4	NC	20

\* Applies only to results above the MDLs.

**Chicago, Illinois 60604**

PROJ. NO. 2001-DR-122104		PROJECT NAME Garfield Alloys				NO. OF CONTAINERS		Activity Code:															
SAMPLERS. (Print Name and Sign) Stephen Wolfe <i>[Signature]</i>																							
STA. NO.	DATE	TIME	COMP.	GRAB	STATION LOCATION			Analyte: TO-14		TAG NUMBERS													
3	6/30/03	01:35		X	14200 Broadway Ave	1	X	REACT	7377	Summa 1													
	Blank					1	X	7378		Trip Blank													
<div>8</div> <div><i>[Signature]</i> Wolfe</div>																							
Relinquished by: (Signature) <i>[Signature]</i> Wolfe		Date / Time 1-2-04 11:45		Received by: (Signature) <i>[Signature]</i> 1/6/03		Ship To: Lockhead Martin / REAC																	
Relinquished by: (Signature) <i>[Signature]</i>		Date / Time 1/6/03 11:05		Received by: (Signature) <i>[Signature]</i> 1/6/03 11:10 AM		ATTN: Jerry Ingram																	
Relinquished by: (Signature)		Date / Time		Received for Laboratory by: (Signature)		Date / Time		Airbill Number 837711835414															
								Chain of Custody Seal Numbers															

Distribution: White - Accompanies Shipment; Pink - Coordinator Field Files; Yellow - Laboratory File

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### APPENDIX B - SUMMA CANISTER DATA

## 1.0 INTRODUCTION

Air samples were collected at the Garfield Alloys Emergency Response site in Cleveland, Ohio on 30 December 2003. A sample and a trip blank were collected in 6-liter passivated Summa canisters. The air samples were transported back to the Environmental Response Team Center (ERTC) facility in Edison, New Jersey. Before sampling, the Summa canisters were cleaned and certified using REAC Standard Operating Procedure (SOP) #1703, "Summa Canister Cleaning Procedures". The samples were analyzed according to REAC SOP #1814, "GC/MS Analysis of Sorbent Tubes and Summa Canisters", using gas chromatography/mass spectrometry (GC/MS).

## 2.0 GC/MS CANISTER PROCEDURES

### 2.1 Sample Pressurization

Before analysis, all canisters were pressurized. A pressurizing train with an in-line pressure gauge accurate to  $\pm 0.1$  pounds per square inch absolute (psia) was used. The gauge and train were purged with nitrogen gas (Ultra High Pure grade) for 5 minutes. The train was connected to the Summa canister and an initial pressure reading was recorded. The Summa canister samples were pressurized with nitrogen and a final pressure reading was recorded. A canister sample was pressurized 2 times the initial reading.

### 2.2 Summa Canister Analysis

Air samples were prepared for GC/MS analysis by cryogenically trapping an aliquot from the Summa canister. The canisters were attached to an Entech Model 7016 Summa Canister Autosampler connected to an Entech Model 7000 Concentrator. Sample analysis was initiated by cooling the first cryotrap, module M-1, to -160 degree Celsius ( $^{\circ}\text{C}$ ). Once M-1 was cooled, an aliquot of sample or standard was cryotrapped on it. This aliquot was transferred to a Tenax trap, M-2, to eliminate most of the water, and then cryofocussed on a third trap, M-3, before injection by direct heating into a Hewlett-Packard 5890 gas chromatography (GC) and 5971A mass selective detector (MSD) running ChemStation software. The cryogenic trap and GC/MS conditions are listed in Table A1.

### 2.3 Calibration and Sample Spiking

A standard mixture (Scott Specialty Gases, Inc. cylinder No. ALM017223) containing twenty-seven (27) target compounds at concentrations of 1.04 to 1.09 parts per million in volume (ppmv) (listed in Table 2) was diluted to a nominal concentration of 20 parts per billion (ppbv) in a Silco-steel passivated canister. An initial calibration was run by varying the volume of the nominal 20 ppbv standard from 50 to 1250 milliliters (mL), equivalent to 1 nanoliter (nL) to 25 nL. A daily standard was analyzed using the nominal 20 ppbv standard at 500 mL (equivalent to 10 nL).

Internal standards, bromochloromethane (BCM), 1,4-difluorobenzene, and chlorobenzene-d5 (Scott Specialty Gases cylinder No. ALM009519), were added to both samples and standards. These standards were diluted from a nominal concentration of 1 ppmv to 100 ppbv in a Silco-steel passivated canister. An aliquot of 100 mL (equivalent to 10 nL) was added to all standards and samples. Instrument performance check standard p-bromofluorobenzene (Scott Specialty Gases cylinder No. ALM057539) were diluted from a nominal concentration of 1 ppmv to 100 ppbv in a Silco-steel passivated canister. An aliquot of 70 mL of BFB (equivalent to 50 nanograms of BFB) was analyzed to validate the mass spectrometer tuning. Standard cylinder I.D. numbers, concentrations, and the quantitation ions are listed in Table A2.

### 2.4 Compound Identification/Quantitation

Target Compounds in the samples were identified and quantitated using ChemStation software. This software was used to tentatively identify and quantitate target compounds using reconstructed and extracted ion chromatogram which were matched with retention time windows. The report format includes the identified compound mass spectra (both raw and background subtracted), quantitation, and qualifier ion chromatogram.

Target compound results were initially reported in nL. The lower calibration standard nominal volume of 1 nL was used as the limit of quantitation (LOQ) for all the target compounds. Target compounds detected at less than twenty-five percent of the LOQ were not reported. The target compound results were calculated in ppbv using the following equation:

$$\text{Concentration(ppbv)} = \frac{\text{Quant Result (nL)} \times 1000}{\text{Undiluted Sample Volume(mL)}}$$

Non-target compounds were identified by a library search of all peaks in a chromatogram. The library search report prints out the sample spectrum along with the ten best library matches and the three best library match spectra. These matches were used along with mass spectral interpretation techniques to tentatively identify the unknowns. Estimated concentrations were calculated based on the total ion response of internal standards in each samples. Non-target compounds with total ion response greater than ten percent of the internal standards' total ion response in each samples were reported; however, non-target compounds appearing in the method blank and compounds, such as siloxanes and carbon dioxide, were not reported.

## 2.5 QA/QC

The following QA/QC procedures were performed for this analysis:

- ▶ The HP 5971A was tuned daily for perfluorotributylamine (PFTBA) to meet abundance criteria for p-bromofluorobenzene as listed in EPA Method 624. Tuning results are included in the QA/QC data section (Appendix B).
- ▶ An initial calibration by automated injection of varying volumes of a 20 ppbv standard were performed on 02 December 2003. All compounds met the relative standard deviations (RSD) of less than 25 %, except for vinyl chloride at 26%.
- ▶ A continuing calibration was performed on 06 January 2004. All compounds met the relative percent difference (RPD) of less than 25 %, except for vinyl chloride at 27.4%.
- ▶ Internal standards were added to all standards and samples. Percent recoveries were calculated against the daily standard, and are listed in Table A6. Recoveries were within 40% to 160% for the internal standards.
- ▶ A method blank were analyzed after the continuing calibration to check for carryover and to ensure that the system was clean.
- ▶ A duplicate was analyzed on sample 14200 Broadway Ave.
- ▶ A set of matrix spike and matrix spike duplicates (MS/MSD) was analyzed on sample 14200 Broadway Ave by spiking the samples with 500 mL of the 20 ppbv standard.

### 3.0 RESULTS

Summa canister target and non target results are listed in Tables A3 and Table A4. All results are reported in ppbv for Summa canister samples and blanks. MS/MSD recoveries are presented in Table A5. Internal standards recoveries are reported in Table A6. The chains-of-custody are in Appendix A. The Summa canister data is in Appendix B.

In Appendix B, the Analysis Log is followed by the calibration package for each day of analysis. The calibration package includes the daily analysis log, canister pressurization log, BFB tune, and initial or continuing calibration quant report. The quant report lists the retention time, quantitation ion, peak area, and concentration in nL. Concentrations listed on this quant reports are generated by using the average response factors of the initial calibration and the response factors of the continuing calibrations.

No target compound were found in sample 14200 Broadway Ave. The sample contained the non-target compounds acetaldehyde, acetone, and butanal, ranging from 4 to 14 ppbv.

A duplicate was analyzed on sample 14200 Broadway Ave. The results for all compounds were very consistent.

The recoveries of MS/MSD on sample 14200 Broadway Ave were ranged from 41 to 113 % and the relative percentage deviations (RPD) ranged from 0 % to 7 %.

### 4.0 DATA ASSESSMENT

The samples were treated with procedures consistent with those described in SOP # 1008, "Operation of Samples Refrigeration Unit and Sample Receiving, Handling and Storage".

The year on the raw data for the acquisition time and quantitation time are incorrectly printed as "104" and "19104". This is related to software problems for year 2004.



TABLE A1 - GC/MS Instrument Conditions

---

B. Preconcentrator Conditions:

M-1 Cryotrap Temperature	: -150°to -160°C
Internal Standard Trap Time	: 1.0 minute
Sample flow	: 150 mL/min
M-1 Cryotrap Desorb Temperature	: 10°to 20°C
M-2 Cryotrap Temperature	: -10°to -20°C
Transfer (M-1 to M-2) Time	: 4.5 minutes
M-2 Cryotrap Desorb Temperature	: 180°C
M-3 Cryotrap Temperature	: -160°C to -180°C
Transfer (M-2 to M-3) Time	: 3.5 minutes
Injection Time	: 2.5 minutes

B. GC/MS Conditions, Sample Analysis:

Initial Temperature	: 40.0°C
Initial Time	: 6.0 minutes
Ramp Rate	: 8.0°C/min
Final Temperature	: 220.0°C
Final Time	: 9.5 minutes
Run Time	: 35.03 minutes
Mass Scan Range:	: 35 to 250 AMU

Column: 0.25 mm x 30 meter Restek RTx-VOA, 3.0 µm film thickness (Restek Corporation)

---

TABLE A2 - Air Toxic Standards (Concentrations and Quantitation Ions)

<u>Compound</u>	<u>Cylinder</u>	<u>Conc. (ppmv)</u>	<u>Quant. Ion</u>
chloromethane	ALM017223	1.03	50
vinyl chloride	ALM017223	1.03	62
chloroethane	ALM017223	1.04	64
trichlorofluoromethane	ALM017223	1.08	101
1,1-dichloroethene	ALM017223	1.02	61
dichloromethane	ALM017223	1.02	49
trans-1,2-dichloroethene	ALM017223	1.02	61
1,1-dichloroethane	ALM017223	1.03	63
cis-1,2-dichloroethene	ALM017223	1.02	61
trichloromethane	ALM017223	1.02	83
1,1,1-trichloroethane	ALM017223	1.02	97
carbon tetrachloride	ALM017223	1.02	117
1,2-dichloroethane	ALM017223	1.02	62
benzene	ALM017223	1.03	78
trichloroethene	ALM017223	1.02	95
bromodichloromethane	ALM017223	1.03	83
dibromomethane	ALM017223	1.03	93
toluene	ALM017223	1.02	91
1,1,2-trichloroethane	ALM017223	1.02	97
tetrachloroethene	ALM017223	1.02	166
ethylbenzene	ALM017223	1.02	91
meta-xylene	ALM017223	1.03	91
ortho-xylene	ALM017223	1.04	91
styrene	ALM017223	1.02	104
1,1,2,2-tetrachloroethane	ALM017223	1.02	83
1,3,5-trimethylbenzene	ALM017223	1.03	120
1,2,4-trimethylbenzene	ALM017223	1.02	105
<u>Internal Standards</u>			
bromochloromethane	ALM040536	1.04	49
1,4-difluorobenzene	ALM040536	1.09	114
chlorobenzene-d5	ALM040536	1.04	117
<u>Instrument Performance Check Standard</u>			
p-bromofluorobenzene (BFB)	ALM057539	1.02	95

**Table A3 - Air Toxic Target Compound Results for Summa Canister Samples**  
**Garfield Alloys Emergency Response Site, Cleveland, OH WA # R1A82001**  
**( concentrations in ppbv )**

Sample Number	Method Blank	REAC4348	REAC4347	REAC4347 Rep
Sample Location	040106-1	Trip Blank	14200 Bway Ave	14200 Bway Ave
Date Sampled	N/A	12/30/03	12/30/03	12/30/03
Date Analyzed	01/06/04	01/06/04	01/06/04	01/06/04
Data File	ERS001	ERS002	ERS003	ERS004
Chloromethane	4 U	4 U	4 U	4 U
Vinyl Chloride	4 U	4 U	4 U	4 U
Chloroethane	4 U	4 U	4 U	4 U
Trichlorofluoromethane	4 U	4 U	4 U	4 U
1,1-Dichloroethene	4 U	4 U	4 U	4 U
Methylene Chloride	4 U	4 U	4 U	4 U
trans-1,2-Dichloroethylene	4 U	4 U	4 U	4 U
1,1-Dichloroethane	4 U	4 U	4 U	4 U
cis-1,2-Dichloroethene	4 U	4 U	4 U	4 U
Trichloromethane	4 U	4 U	4 U	4 U
1,1,1-Trichloroethane	4 U	4 U	4 U	4 U
Carbon Tetrachloride	4 U	4 U	4 U	4 U
1,2-Dichloroethane	4 U	4 U	4 U	4 U
Benzene	4 U	4 U	4 U	4 U
Trichloroethylene	4 U	4 U	4 U	4 U
Bromodichloromethane	4 U	4 U	4 U	4 U
Dibromomethane	4 U	4 U	4 U	4 U
Toluene	4 U	4 U	4 U	4 U
1,1,2-Trichloroethane	4 U	4 U	4 U	4 U
Tetrachloroethylene	4 U	4 U	4 U	4 U
Ethylbenzene	4 U	4 U	4 U	4 U
m & p-Xylenes	4 U	4 U	4 U	4 U
o-Xylene	4 U	4 U	4 U	4 U
Styrene	4 U	4 U	4 U	4 U
1,1,2,2-Tetrachloroethane	4 U	4 U	4 U	4 U
1,3,5-Trimethylbenzene	4 U	4 U	4 U	4 U
1,2,4-Trimethylbenzene	4 U	4 U	4 U	4 U
Pressurized Sample Volume (mL)	250	250	500	500
Initial Pressure (psia)	N/A	N/A	14.2	14.2
Final Pressure (psia)	N/A	N/A	28.4	28.4
Quantitation Limit (ppbv)	4	4	4	4

A - Assumed volume for Blanks  
B - <3 times Method Blank value  
C - Compound Calibration >25% RSD  
D - Compound Calibration Check >25% RPD  
E - Concentration exceeded calibration limit (25nL)  
J - Below 1.00 nL Quantitation Limit  
U - Not Detected  
N/A - Not Applicable  
ppbv - Parts per billion by volume

Table A4- Air Toxic Non-target Compounds  
Summa Canister Sample Results

Garfield Alloys ER Site, Cleveland, OH, R1A82001

Sample Number: Method Blank Page 1 of 4  
Sample Location: 040106-1  
Sample Volume (mL): 250  
Date Sampled: N/A  
Date Analyzed: 01/06/04 Initial Pressure N/A  
Data File: ERS0001 Final Pressure N/A

Compound Name	Retention Time	Area	Concentration(ppbv)
---------------	----------------	------	---------------------

No non-targets were found.
----------------------------

Bromochloromethane (13.1)

1,4-Difluorobenzene (18.0)

Chlorobenzene-D5 (40.0)

\* - Below 4 ppbv Limit of Quantitation

N/A - Not Applicable

Table A4- Air Toxic Non-target Compounds  
Summa Canister Sample Results

Garfield Alloys ER Site, Cleveland, OH, R1A82001

Sample Number: REAC4348 Page 2 of 4  
Sample Location: Trip Blank  
Sample Volume (mL): 250  
Date Sampled: 12/30/03  
Date Analyzed: 01/06/04 Initial Pressure N/A  
Data File: ERS0002 Final Pressure N/A

Compound Name	Retention Time	Area	Concentration(ppbv)
---------------	----------------	------	---------------------

No non-targets were found.
----------------------------

Bromochloromethane (13.1)

1,4-Difluorobenzene (18.0)

Chlorobenzene-D5 (40.0)

\* - Below 4 ppbv Limit of Quantitation

N/A - Not Applicable

Table A4- Air Toxic Non-target Compounds  
Summa Canister Sample Results

Garfield Alloys ER Site, Cleveland, OH, R1A82001

Sample Number: REAC4347 Page 3 of 4  
Sample Location: 14200 Bway Ave  
Sample Volume (mL): 500  
Date Sampled: 12/30/03  
Date Analyzed: 01/06/04 Initial Pressure 14.2  
Data File: ERS0003 Final Pressure 28.4

Compound Name	Retention Time	Area	Concentration(ppbv)
unknown	2.983	392666	4 *
acetaldehyde	4.161	1002981	9
acetone	7.341	1594189	14
butanal	11.094	434590	4 *
2-butanone	11.285	431892	4 *

Bromochloromethane (13.1)

1,4-Difluorobenzene (18.0)

Chlorobenzene-D5 (40.0)

\* - Below 4 ppbv Limit of Quantitation

N/A - Not Applicable

**Table A4- Air Toxic Non-target Compounds  
Summa Canister Sample Results**

**Garfield Alloys ER Site, Cleveland, OH, R1A82001**

Sample Number: **REAC4347 Rep** Page 4 of 4  
 Sample Location: **14200 Bway Ave**  
 Sample Volume (mL): **500**  
 Date Sampled: **12/30/03**  
 Date Analyzed: **01/06/04** Initial Pressure **14.2**  
 Data File: **ERS0004** Final Pressure **28.4**

Compound Name	Retention Time	Area	Concentration(ppbv)
unknown	2.983	381252	3 *
acetaldehyde	4.153	1001022	9
acetone	7.333	1562925	14
butanal	11.086	444304	4 *
2-butanone	11.285	445461	4

Bromochloromethane (13.1)

1,4-Difluorobenzene (18.0)

Chlorobenzene-D5 (40.0)

\* - Below 4 ppbv Limit of Quantitation

N/A - Not Applicable

**Table A5 - Air Toxic MS/MSD Recovery Summary for Summa Canister Samples**  
**Garfield Alloys Emergency Response Site, Cleveland, OH WA # R1A82001**  
**( concentrations in nL )**

Sample Number		REAC4347	REAC4347 MS		REAC4347 MSD		
Sample Location		14200 Bway Ave	14200 Bway Ave		14200 Bway Ave		
Date Sampled		12/30/03	12/30/03		12/30/03		
Date Analyzed	Spike	01/06/04	01/06/04	%	01/06/04	%	
Data File	Amount	ERS003	ERS005	Recovery	ERS006	Recovery	RPD
Chloromethane	10.3	0.11	10.64	102	10.54	101	1
Vinyl Chloride	10.3	0.00	10.56	103	10.41	101	1
Chloroethane	10.4	0.00	4.28	41	4.25	41	1
Trichlorofluoromethane	10.8	0.00	7.30	68	7.39	68	1
1,1-Dichloroethene	10.2	0.00	6.53	64	6.49	64	1
Methylene Chloride	10.2	0.00	4.84	47	4.77	47	1
trans-1,2-Dichloroethene	10.2	0.00	6.00	59	6.45	63	7
1,1-Dichloroethane	10.3	0.00	9.23	90	9.25	90	0
cis-1,2-Dichloroethene	10.2	0.00	9.75	96	9.74	95	0.1
Trichloromethane	10.2	0.00	9.75	96	9.57	94	2
1,1,1-Trichloroethane	10.2	0.00	9.63	94	9.66	95	0.3
Carbon Tetrachloride	10.2	0.00	9.50	93	9.49	93	0
1,2-Dichloroethane	10.2	0.00	10.48	103	10.23	100	2
Benzene	10.3	0.00	9.49	92	9.45	92	0
Trichloroethylene	10.2	0.00	9.33	91	9.30	91	0
Bromodichloromethane	10.3	0.00	10.09	98	10.02	97	1
Dibromomethane	10.3	0.00	10.31	100	10.27	100	0
Toluene	10.2	0.00	10.06	99	10.00	98	1
1,1,2-Trichloroethane	10.2	0.00	10.50	103	10.50	103	0
Tetrachloroethylene	10.2	0.00	9.93	97	9.83	96	1
Ethylbenzene	10.2	0.00	9.70	95	9.53	93	2
meta & para-Xylenes	10.3	0.00	9.79	95	9.70	94	1
ortho-Xylene	10.4	0.00	9.79	94	9.73	94	1
Styrene	10.2	0.00	10.10	99	10.16	100	1
1,1,2,2-Tetrachloroethane	10.2	0.00	11.38	112	11.52	113	1
1,3,5-trimethylbenzene	10.3	0.00	9.62	93	9.59	93	0
1,2,4-trimethylbenzene	10.2	0.00	10.02	98	10.16	100	1

nL - Nanoliter



**Table A6 - Air Toxic Internal Standards Recovery Summary**  
**Summa Canister Samples**

**Garfield Alloys Emergency Response Site, Cleveland, OH WA# R1A82001**

Internal Standards		Bromochloromethane	% Recovery	1,4-Difluorobenzene	% Recovery	Chlorobenzene-d <sub>5</sub>	% Recovery
Area of continuing calibration		1167437	100%	3601577	100%	2785528	100%
Allowable Maximum area (160%)		1867899	160%	5762523	160%	4456845	160%
Allowable Minimum area (40%)		466975	40%	1440631	40%	1114211	40%
Area of samples							
ERS001	Method Blank 040106-1	1104316	95%	3530706	98%	2654081	95%
ERS002	REAC4348 Trip Blank	1113347	95%	3552388	99%	2664861	96%
ERS003	REAC4347 14200 Bway Ave	1073471	92%	3385812	94%	2626655	94%
ERS004	REAC4347 14200 Bway Ave Rep	1069576	92%	3350430	93%	2620976	94%
ERS005	REAC4347 14200 Bway Ave MS	1096374	94%	3237797	90%	2617428	94%
ERS006	REAC4347 14200 Bway Ave MS	1105375	95%	3233503	90%	2613427	94%

Attachment H

REAC Air Sample Results

ANALYTICAL REPORT

Prepared by  
LOCKHEED MARTIN, Inc.

Garfield Alloys Fire ER Site  
Garfield Heights, OH

February 2004

EPA Work Assignment No. 82001  
LOCKHEED MARTIN Work Order R1A 82001  
EPA Contract No. 68-C99-223

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Appendix will be furnished on request.	

## Introduction

REAC in response to WA # 82001, provided analytical support for environmental samples collected from Garfield Alloys Fire ER Site, located in Garfield Heights, OH as described in the following table. The support also included QA/QC, data review, and preparation of an analytical report containing a summary of the analytical methods, the results, and the QA/QC results.

The samples were treated with procedures consistent with those specified in SOP #1008.

COC #	Number of Samples	Sampling Date	Date Received	Matrix	Analysis	Laboratory	Data Package
06126	14	12/30/03	12/31/03	Air	TAL Metals	REAC	N 019

## Case Narrative

The data in this report have been validated to two significant figures. Any other representation of the data is the responsibility of the user.

### TAL Metals in Air Package N 019

The original request on the chain of custody was for magnesium. At the request of the Task Leader, the samples were analyzed for TAL metals. The results of the magnesium analysis have been given in a previous report.

The field and trip blanks contained 0.15 µg/filter chromium and 0.059 µg/filter chromium, respectively after media blank subtraction. The chromium results for samples 14146, 14147, 14148, 14151 and 14152 should be regarded as not detected (U).

## Summary of Abbreviations

AA	Atomic Absorption
B	The analyte was found in the blank
BFB	Bromofluorobenzene
C	Centigrade
cont.	Continued
D	(Surrogate Table) this value is from a diluted sample and was not calculated (Result Table) this result was obtained from a diluted sample
Dioxin and/or	
PCDD and PCDF	denotes Polychlorinated Dibenzo-p-dioxins and Polychlorinated Dibenzofurans
CLP	Contract Laboratory Protocol
COC	Chain of Custody
CONC	Concentration
CRDL	Contract Required Detection Limit
CRQL	Contract Required Quantitation Limit
DFTPP	Decafluorotriphenylphosphine
DL	Detection Limit
E	The value is greater than the highest linear standard and is estimated
EMPC	Estimated maximum possible concentration
ICAP	Inductively Coupled Argon Plasma
ISTD	Internal Standard
J	The value is below the method detection limit and is estimated
LCS	Laboratory Control Sample
LCSD	Laboratory Control Sample Duplicate
MDL	Method Detection Limit
MI	Matrix Interference
MS (BS)	Matrix Spike (Blank Spike)
MSD (BSD)	Matrix Spike Duplicate (Blank Spike Duplicate)
MW	Molecular Weight
NA	either Not Applicable or Not Available
NC	Not Calculated
NR	Not Requested
NS	Not Spiked
% D	Percent Difference
% REC	Percent Recovery
PPB	Parts per billion
PPBV	Parts per billion by volume
PPMV	Parts per million by volume
PQL	Practical Quantitation Limit
QA/QC	Quality Assurance/Quality Control
QL	Quantitation Limit
RPD	Relative Percent Difference
RSD	Relative Standard Deviation
SIM	Selected Ion Monitoring
TCLP	Toxicity Characteristic Leaching Procedure
TIC	Tentatively Identified Compound
U	Denotes not detected
W	Weathered analyte; Aroclor pattern displays a degradation of earlier eluting peaks
m <sup>3</sup>	cubic meter      kg      kilogram      µg      microgram
L	liter      g      gram      pg      picogram
mL	milliliter      mg      milligram      ng      nanogram
µL	microliter
*	denotes a value that exceeds the acceptable QC limit
	Abbreviations that are specific to a particular table are explained in footnotes on that table

Revision 7/16/03

## Analytical Procedure for Metals in Air

### Sample Preparation

Each cassette filter holder was carefully opened, and the filter sample was transferred to a clean 50 mL beaker and prepared according to ERTC/REAC SOP #1813, *Analysis of Metals in Air with a Modified NIOSH 7300 Method*. The samples were mixed with 5-mL concentrated nitric acid and heated using an adjustable heating device, capable of maintaining a temperature of 90 - 95° C, until the volume was reduced to approximately 0.5 mL. After digestion, the samples were allowed to cool to room temperature, transferred to 25 mL volumetric flasks, diluted to volume with dilution acid, and analyzed for all metals, except mercury, according to SOP #1811 *Determination of Metals by Inductively Coupled Plasma (ICP) Methods*.

One blank spike (BS) and one blank spike duplicate (BSD) sample (prepared using blank filters) were processed for each analytical batch of samples.

### Analysis and Calculations

The ICP instrument was calibrated and operated according to SOP # 1811, and the manufacturer's operating instructions. After calibration, the initial calibration verification (ICV), initial calibration blank (ICB), and QC check standards were run to verify proper calibration. The continuing calibration verification (CCV) and continuing calibration blank (CCB) standards were run after every ten samples to assure proper operation during sample analysis.

The metal concentrations in solution were read directly from the read-out system of the ICP instrument. The concentration of metal ( $\mu\text{g}/\text{m}^3$ ) in the air volume sampled was:

$$\mu\text{g}/\text{m}^3 = 1000 \times [(W - \text{BLK}) / V]$$

where:

W	= amount of metal on each filter ( $\mu\text{g}$ )
	= A x FV
A	= concentration of metal in the sample ( $\mu\text{g}/\text{L}$ )
	= B x CF x DF
B	= instrument read-out ( $\mu\text{g}/\text{L}$ for ICP)
CF	= conversion factor (1.00 for $\mu\text{g}/\text{L}$ )
DF	= dilution factor for diluted sample (1.00 with no sample dilution)
FV	= final solution volume (L)
BLK	= amount of metal in the media blank ( $\mu\text{g}$ )
V	= Volume of Air sampled (L)

NOTE: BLK is the average of the media blank results (minimum 3). For blank values <MDL, substitute zero (0) for the raw data prior to calculating the average. The calculated average is then subtracted from each sample even if the average is < MDL.

For samples that required dilution to be within the instrument calibration range, DF is given by:

$$\text{DF} = (\text{C} + \text{B}) / \text{C}$$

where:

B	= amount of acid blank used for dilution (mL)
C	= sample aliquot (mL)

The results are listed in Table 1.1.  
Revision date: 04/24/2003

Table 1.1 Results of the Analysis for Metals in Air  
WA # 82-001 Garfield Alloy Site

Client ID Location Air Volume (L)		Media Blank #1 Lab		Media Blank#2 Lab		Media Blank#3 Lab		14153 Field Blank 0		14154 Trip Blank 0		14155 Lot Blank 0	
Parameter	Analysis Method	Conc µg/filter	MDL µg/filter	Conc µg/filter	MDL µg/filter	Conc µg/filter	MDL µg/filter	Conc µg/filter	MDL µg/filter	Conc µg/filter	MDL µg/filter	Conc µg/filter	MDL µg/filter
Aluminum	ICAP	U	0.63	U	0.63	U	0.63	U	0.63	U	0.63	U	0.63
Antimony	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Arsenic	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Barium	ICAP	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05
Beryllium	ICAP	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05
Cadmium	ICAP	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05
Calcium	ICAP	4.0	2.5	4.0	2.5	3.7	2.5	U	2.5	U	2.5	U	2.5
Chromium	ICAP	0.87	0.05	0.69	0.05	0.73	0.05	0.15	0.05	0.059	0.05	U	0.05
Cobalt	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Copper	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Iron	ICAP	0.28	0.25	0.32	0.25	0.25	0.25	U	0.25	U	0.25	U	0.25
Lead	ICAP	0.22	0.075	0.27	0.075	0.089	0.075	U	0.075	U	0.075	U	0.075
Magnesium	ICAP	U	5.0	U	5.0	U	5.0	U	5.0	U	5.0	U	5.0
Manganese	ICAP	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05
Nickel	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Potassium	ICAP	U	5.0	U	5.0	U	5.0	U	5.0	U	5.0	U	5.0
Selenium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Silver	ICAP	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05	U	0.05
Sodium	ICAP	U	25	U	25	U	25	U	25	U	25	U	25
Thallium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Vanadium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Zinc	ICAP	0.79	0.13	0.55	0.13	0.67	0.13	U	0.13	U	0.13	U	0.13



Table 1.1 (cont.) Results of the Analysis for Metals in Air  
WA # 82-001 Garfield Alloy site

Client ID Location		14141 Station#10 (DW) 960		14142 Station#11 (DW) 916		14143 Station#12 (Source) 960		14144 Station#13 (UW) 956		14145 Station#7 960		14146 Station#8 960	
Air Volume (L)													
Parameter	Analysis Method	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³
Aluminum	ICAP	U	0.65	U	0.68	0.68	0.65	U	0.65	U	0.65	U	0.65
Antimony	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Arsenic	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Barium	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	U	0.052
Beryllium	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	U	0.052
Cadmium	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	U	0.052
Calcium	ICAP	U	2.6	U	2.7	U	2.6	U	2.6	U	2.6	U	2.6
Chromium	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	0.24	0.052
Cobalt	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Copper	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Iron	ICAP	0.76	0.26	2.6	0.27	1.4	0.26	0.95	0.26	0.36	0.26	0.29	0.26
Lead	ICAP	U	0.078	U	0.082	0.70	0.078	U	0.078	U	0.078	U	0.078
Magnesium	ICAP	U	5.2	U	5.5	U	5.2	U	5.2	U	5.2	U	5.2
Manganese	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	0.057	0.052
Nickel	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Potassium	ICAP	U	5.2	U	5.5	U	5.2	U	5.2	U	5.2	U	5.2
Selenium	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Silver	ICAP	U	0.052	U	0.055	U	0.052	U	0.052	U	0.052	U	0.052
Sodium	ICAP	U	26	U	27	U	26	U	26	U	26	U	26
Thallium	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Vanadium	ICAP	U	0.13	U	0.14	U	0.13	U	0.13	U	0.13	U	0.13
Zinc	ICAP	U	0.13	U	0.14	0.25	0.13	U	0.13	U	0.13	U	0.13

Table 1.1 (cont.) Results of the Analysis for Metals in Air  
WA # 82-001 Garfield Alloy site

Client ID Location Air Volume (L)		14147 Station#6 960		14148 Station#5 960		14149 Station#4 960		14151 Station#2 960		14152 Station#1 960	
Parameter	Analysis Method	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³	Conc µg/m³	MDL µg/m³
Aluminum	ICAP	U	0.65	0.75	0.65	U	0.65	U	0.65	U	0.65
Antimony	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Arsenic	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Barium	ICAP	U	0.052	U	0.052	U	0.052	U	0.052	U	0.052
Beryllium	ICAP	U	0.052	U	0.052	U	0.052	U	0.052	U	0.052
Cadmium	ICAP	U	0.052	U	0.052	U	0.052	U	0.052	U	0.052
Calcium	ICAP	U	2.6	U	2.6	U	2.6	U	2.6	U	2.6
Chromium	ICAP	0.10	0.052	0.15	0.052	U	0.052	0.078	0.052	0.11	0.052
Cobalt	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Copper	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Iron	ICAP	U	0.26	0.74	0.26	0.70	0.26	0.47	0.26	0.38	0.26
Lead	ICAP	U	0.078	U	0.078	U	0.078	U	0.078	0.084	0.078
Magnesium	ICAP	U	5.2	U	5.2	U	5.2	U	5.2	U	5.2
Manganese	ICAP	U	0.052	U	0.052	U	0.052	U	0.052	U	0.052
Nickel	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Potassium	ICAP	U	5.2	U	5.2	U	5.2	U	5.2	U	5.2
Selenium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Silver	ICAP	U	0.052	U	0.052	U	0.052	U	0.052	U	0.052
Sodium	ICAP	U	26	U	26	U	26	U	26	U	26
Thallium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Vanadium	ICAP	U	0.13	U	0.13	U	0.13	U	0.13	U	0.13
Zinc	ICAP	U	0.13	0.17	0.13	U	0.13	0.24	0.13	U	0.13

QA/QC for TAL Metals in Air

Results of the BS/BSD Analysis for TAL Metals in Air

A blank was spiked and analyzed in duplicate as a blank spike/blank spike duplicate (BS/BSD). The percent recoveries, listed in Table 2.1, ranged from 81 to 110 and all forty-six values were within the acceptable QC limits. The relative percent differences, also listed in Table 2.1, ranged from 0 (zero) to 16 and all twenty-three values were within the acceptable QC limits.

Table 2.1 Results of the BS/BSD Analysis for Metals in Air  
WA # 82-001 Garfield Alloy Site

Sample ID: Metal	BS/BSD BS Spike Added µg/filter	BS Conc µg/filter	BS % Rec	BSD Spike Added µg/filter	BSD Conc µg/filter	BSD % Rec	RPD	Recommended QC Limits % Rec	RPD
Aluminum	10	9.8	98	10	10.3	103	5	75-125	20
Antimony	1.00	0.893	89	1.00	0.98	98	9	75-125	20
Arsenic	1.00	0.981	98	1.00	1.05	105	7	75-125	20
Barium	2.50	2.46	98	2.50	2.54	102	3	75-125	20
Beryllium	2.50	2.43	97	2.50	2.52	101	4	75-125	20
Cadmium	2.50	2.44	98	2.50	2.55	102	4	75-125	20
Calcium	50	47.7	95	50	49.8	100	4	75-125	20
Chromium	2.50	2.62	105	2.50	2.63	105	0	75-125	20
Cobalt	2.50	2.47	99	2.50	2.59	104	5	75-125	20
Copper	2.50	2.66	106	2.50	2.76	110	4	75-125	20
Iron 2599	10	10.1	101	10	10.6	106	5	75-125	20
Iron 2714	10	9.8	98	10	10.6	106	8	75-125	20
Lead	1.00	0.81	81	1.00	0.95	95	16	75-125	20
Magnesium	50	48	96	50	50.4	101	5	75-125	20
Manganese	2.50	2.51	100	2.50	2.6	104	4	75-125	20
Nickel	2.50	2.43	97	2.50	2.56	102	5	75-125	20
Potassium	50	46.6	93	50	50.2	100	7	75-125	20
Selenium	1.00	0.988	99	1.00	0.99	99	0	75-125	20
Silver	2.50	2.2	88	2.50	2.36	94	7	75-125	20
Sodium	200	188	94	200	199	100	6	75-125	20
Thallium	1.00	0.964	96	1.00	1.02	102	6	75-125	20
Vanadium	2.50	2.42	97	2.50	2.52	101	4	75-125	20
Zinc	2.50	2.42	97	2.50	2.63	105	8	75-125	20

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### Analyses Requested

Items/Reason	Relinquished by	Date	Received by	Date	Time	Items/Reason	Relinquished by	Date	Received by	Date	Time
All/Analysis	M. Iresparakos	12-31-03	Joe R	12/31/03	1:33	411/750	Joe R	12/31/03	C. Hason	12/31/03	1:50